

ASTROCHEMISTRY : FROM STARS TO PLANETS

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Table

1. Why molecules?

- It all begins with nuclei
- Relevance

2. Where molecules?

- Hot to cold
- Dense to thin
- Ionized, Plasmas, PDRs

3. When molecules?

- Early universe
- Disks
- Planets

4. How to detect?

- Rotation
- IR and other

5. How to model?

- Qualitative/Quantitative
- Ab initio

6. Conclusion

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6. Conclusion

Chemistry = molecules

ISM

Table 2
List of Detected Interstellar Molecules with Two to Seven Atoms, Categorized by Number of Atoms, and Vertically Ordered by Detection Year

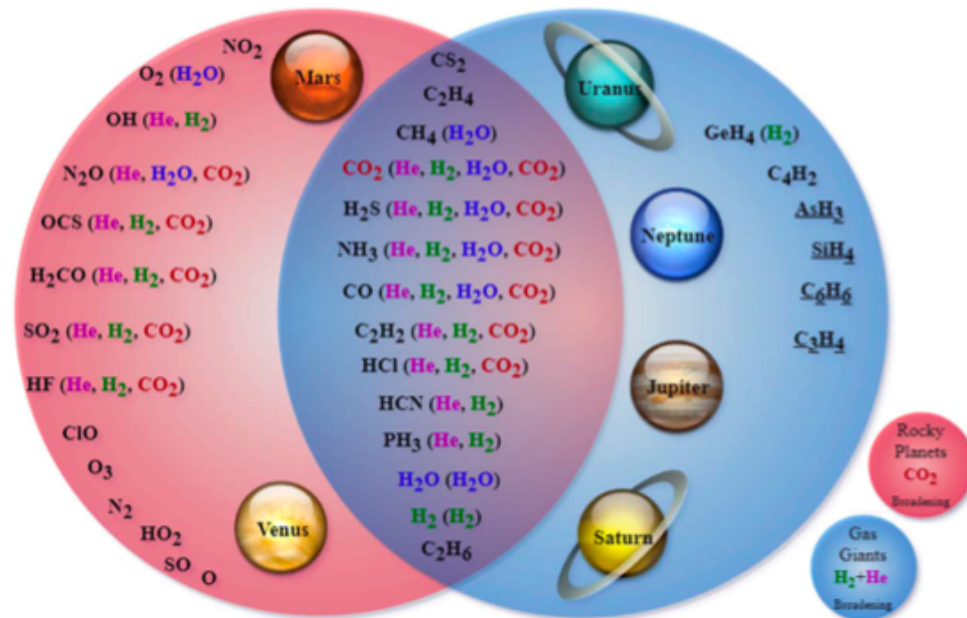
2 Atoms		3 Atoms		4 Atoms		5 Atoms		6 Atoms	7 Atoms
CH	NH	H ₂ O	MgCN	NH ₃	SiC ₃	HC ₃ N	C ₄ H ⁻	CH ₃ OH	CH ₃ CHO
CN	SiN	HCO ⁺	H ₃ ⁺	H ₂ CO	CH ₃	HCOOH	CNCHO	CH ₃ CN	CH ₃ CCH
CH ⁺	SO ⁺	HCN	SiCN	HNCO	C ₃ N ⁻	CH ₂ NH	HNCNH	NH ₂ CHO	CH ₃ NH ₂
OH	CO ⁺	OCS	AINC	H ₂ CS	PH ₃	NH ₂ CN	CH ₃ O	CH ₃ SH	CH ₂ CHCN
CO	HF	HNC	SiNC	C ₂ H ₂	HCNO	H ₂ CCO	NH ₃ D ⁺	C ₂ H ₄	HC ₃ N
H ₂	N ₂	H ₂ S	HCP	C ₃ N	HOCN	C ₄ H	H ₂ NCO ⁺	C ₃ H	C ₆ H
SiO	CF ⁺	N ₂ H ⁺	CCP	HNCS	HSCN	SiH ₄	NCCNH ⁺	CH ₃ CN	c-C ₂ H ₄ O
CS	PO	C ₂ H	AlOH	HOCO ⁺	HOOH	c-C ₃ H ₂	CH ₃ Cl	HC ₂ CHO	CH ₂ CHOH
SO	O ₂	SO ₂	H ₂ O ⁺	C ₃ O	1-C ₃ H ⁺	CH ₂ CN	MgC ₃ N	H ₂ C ₄	C ₆ H ⁻
SiS	AlO	HCO	H ₂ Cl ⁺	1-C ₃ H	HMgNC	C ₅	HC ₃ O ⁺	C ₅ S	CH ₃ NCO
NS	CN ⁻	HNO	KCN	HCNH ⁺	HCCO	SiC ₄	NH ₂ OH	HC ₃ NH ⁺	HC ₃ O
C ₂	OH ⁺	HCS ⁺	FeCN	H ₃ O ⁺	CNCN	H ₂ CCC	HC ₃ S ⁺	C ₃ N	HOCH ₂ CN
NO	SH ⁺	HOC ⁺	HO ₂	C ₃ S	HONO	CH ₄	H ₂ CSS	HC ₄ H	HC ₄ NC
HCl	HCl ⁺	SiC ₂	TiO ₂	c-C ₃ H	MgCCH	HCCNC	C ₄ S	HC ₄ N	H ₃ HNH
NaCl	SH	C ₂ S	CCN	HC ₂ N	HCCS	HNCCC	CHOSH	c-H ₂ C ₃ O	c-C ₃ HCCH
AlCl	TiO	C ₃	SiCSi	H ₂ CN		H ₂ COH ⁺		CH ₂ CNH	
KCl	ArH ⁺	CO ₂	S ₂ H					C ₅ N ⁻	
AlF	NS ⁺	CH ₂	HCS					HNCHCN	
PN	HeH ⁺	C ₂ O	HSC					SiH ₃ CN	
SiC	VO	MgNC	NCO					MgC ₄ H	
CP		NH ₂	CaNC					CH ₃ CO ⁺	
		NaCN	NCS					H ₂ CCCS	
		N ₂ O						CH ₂ CCH	

Table 3

List of Detected Interstellar Molecules with Eight or More Atoms, Categorized by Number of Atoms, and Vertically Ordered by Detection Year

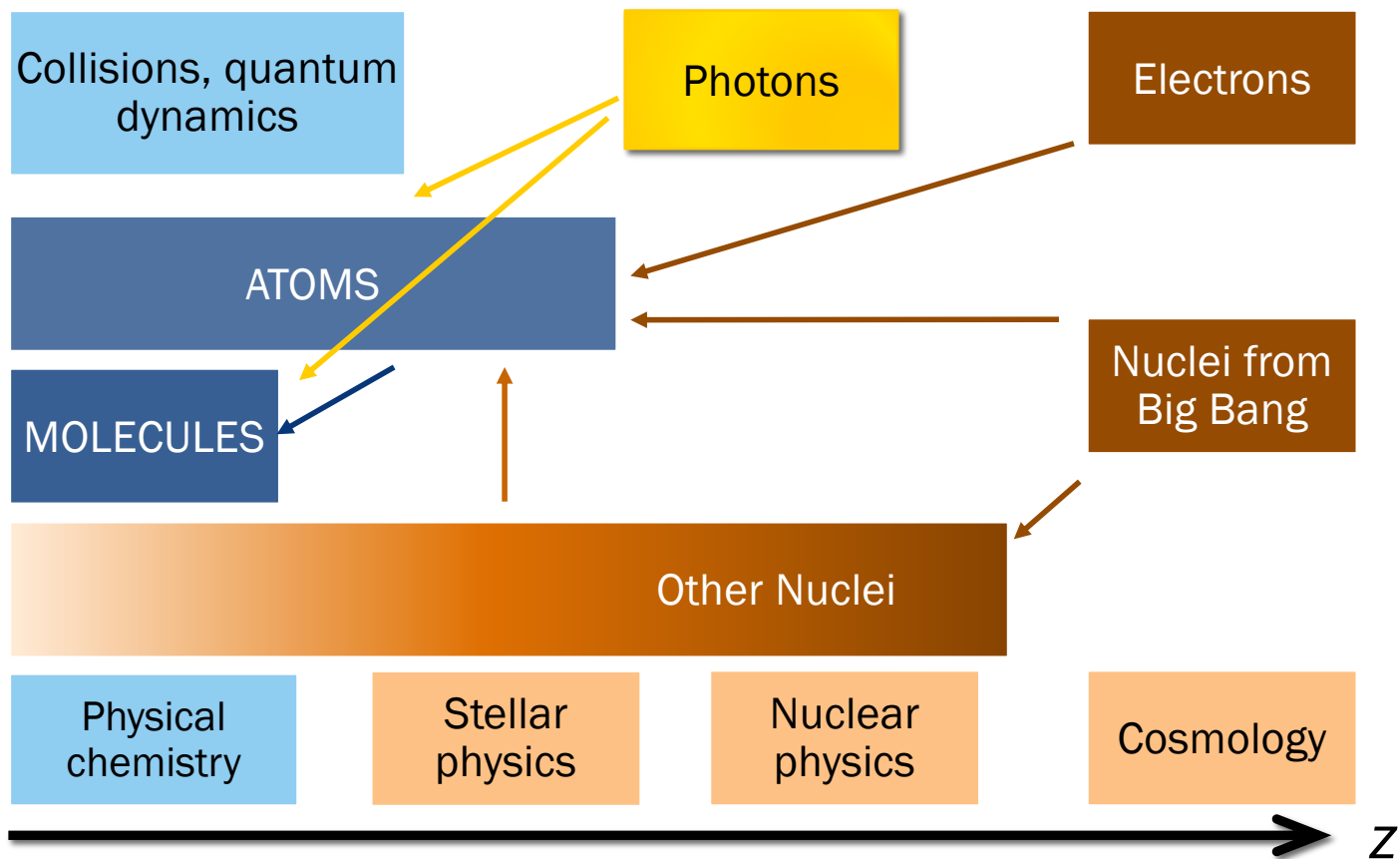
8 Atoms	9 Atoms	10 Atoms	11 Atoms	12 Atoms	13 Atoms	PAHs	Fullerenes
HCOOCH ₃	CH ₃ OCH ₃	CH ₃ COCH ₃	HC ₉ N	C ₆ H ₆	C ₆ H ₅ CN	1-C ₁₀ H ₇ CN	C ₆₀
CH ₃ C ₃ N	CH ₃ CH ₂ OH	HOCH ₂ CH ₂ OH	CH ₃ C ₆ H	n-C ₃ H ₇ CN	HC ₁₁ N	2-C ₁₀ H ₇ CN	C ₆₀ ⁺
C ₇ H	CH ₃ CH ₂ CN	CH ₃ CH ₂ CHO	C ₂ H ₃ OCHO	i-C ₃ H ₇ CN		C ₉ H ₈	C ₇₀
CH ₃ COOH	HC ₇ N	CH ₃ C ₃ N	CH ₃ COOCH ₃	1-C ₃ H ₅ CN			
H ₂ C ₆	CH ₃ C ₄ H	CH ₃ CHCH ₂ O	CH ₃ COCH ₂ OH	2-C ₃ H ₅ CN			
CH ₂ OHCHO	C ₈ H	CH ₃ CH ₂ OH	C ₅ H ₆				
HC ₆ H	CH ₃ CONH ₂						
CH ₂ CHCHO	C ₈ H ⁻						
CH ₂ CCHCN	CH ₂ CHCH ₃						
NH ₂ CH ₂ CN	CH ₃ CH ₂ SH						
CH ₃ CHNH	HC ₇ O						
CH ₃ SiH ₃	CH ₃ NHCHO						
NH ₂ CONH ₂	H ₂ CCCHCCH						
HCCCH ₂ CN	HCCCHCHCN						
CH ₂ CHCCH	H ₂ CCHC ₃ N						

Chemistry = molecules



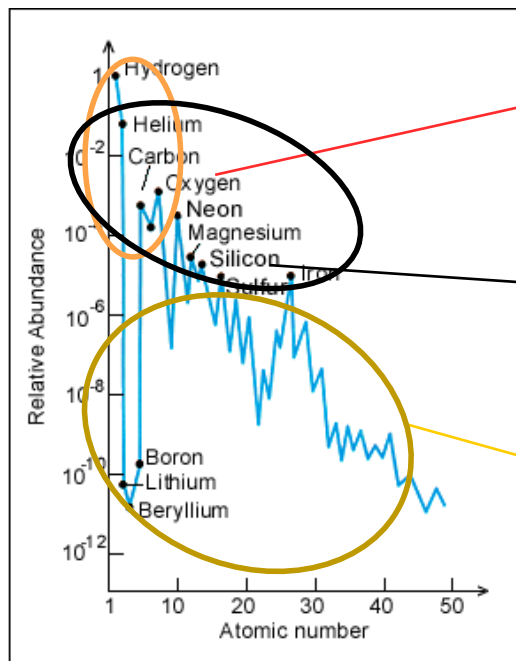
*Planetary atmospheres
main gases and
prominent molecules*

A Hierarchy of Ingredients



NUCLEAR PHYSICS

Elemental abundances in the Milky Way



Organic chemistry
main elements:
 $H \gg O \sim C > N$

He + Some
heavier elements

Little P, Alkalines,
Halogens
VERY little Li, Be, B

FIGURE 3.46 — A summary of the cosmic abundances of the elements and their isotopes.

MOLECULAR PHYSICS

Why looking for molecules ?

- **Molecules** display a nearly infinite of variety:
 - from the simplest, H_2 , the transition-elements containing molecules to the molecules of life, and the polymers, the PAH's .
 - All categories have been observed in extraterrestrial environments, from meteorites to the edge of the Universe.
- **Molecules** (and in particular isotopologues) are a beacon for history

- **Molecules**

are seen in nearly any environment: binding energies from a few eV \rightarrow to a few Kelvin

- **In Astrophysics,**

The abundance of nuclei is determinantal for the abundances of the various molecules: interplay of GeV/MeV to eV/ μ eV physics (15 orders of magnitude), interplay of high and low energy astrophysics, including super novae

- Measures of fundamental constants history

Tool	Some results
Thermodynamics and kinetics	Out of equilibrium situations, timescales
Spectroscopy	Fundamental constants, identification, abundances
Spectral lineshapes	Opacities, atmospheres (planets and cool stars)
Isotopic compositions	Detailed history, past nuclear physics
Molecular presence and abundances	Physical and chemical characterization, atmospheric composition
Mass spectrometry (probes)	Up to large molecules, polymers

Why is Astrochemistry both an ancient and a modern problem?

Is it good old spectroscopy again, in a new guise? Yes, and difficult.

- Think of the hyperfine structure of ND_2H , the rotational-torsional spectroscopy of $\text{CH}_3\text{-O-CH}_3$, to the ro-vibrational spectroscopy of exotic ions $(\text{H}_3\text{CH}_2\text{CN})^+$ or radicals (CHNH) . All these species are known, and some have been discovered by astrophysical spectroscopy.
- **Spectroscopy** ranges from GHz to THz and beyond (cm \rightarrow sub-mm, FIR, IR). We have all the instruments, and everything in the spectral line is significant. “Every photon counts”
- **Spectral line shapes are not in equilibrium**, they are broadened by the Doppler effect and by collisions, they are heterogeneous and optically thick.
- **Spectra of highly excited atoms** is very difficult to model

Why is Astrochemistry both an ancient and a modern problem?

- ***It's also about:*** finding a flower in a meadow of weeds. Weeds are common and hardy. Flowers are rare, their signals weak, but what an achievement to find **urea $\text{O}=\text{C}(\text{NH}_2)_2$** - the first biomolecule synthesized in the 19th century, which proved that chemistry was universal- sugars and amino acids (these, in meteorites, and perhaps in comets).
- ***It's also about:*** characterizing the atmospheres of other stars and their planets similar to Neptune, or superbly strange like the Trappist system: 5 planets - IR star. Study meteorology there and one day find bio-signed molecules.

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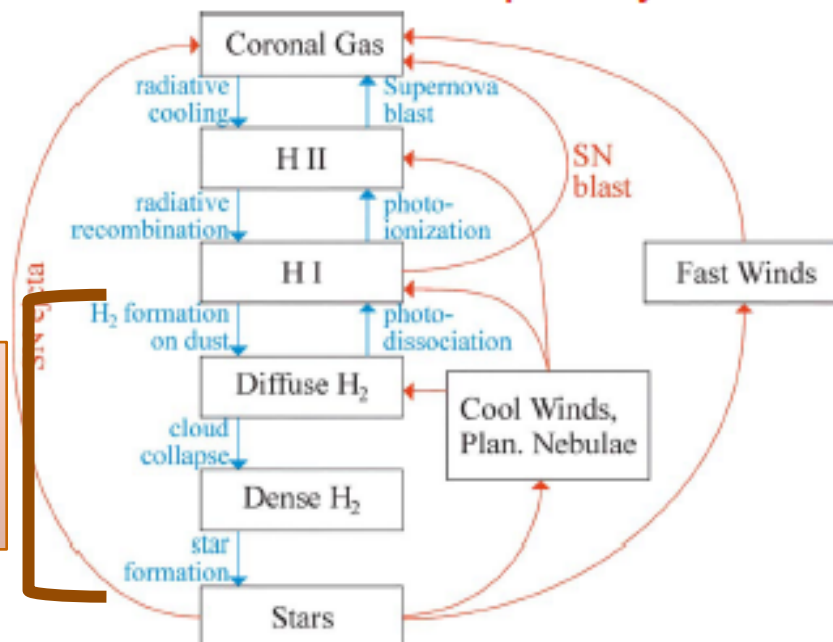
6. Conclusion

A list (not exhaustive...)

*Not mentioned:
PAH's;
Intergalactic*

Object	Conditions
PDR	gradients in ionization temperature. Photophysics dominant.
HI regions	Very low density, equilibrium very slow to reach
Molecular ISM + grains	Cold, some molecules freeze out (deplete) on grain surfaces. Gas and surface chemistry
YSO	Factories of COM's
Dying stars	Factories of COM's
Disks	Tracers of future history. Gradients in space, turbulence, Keplerian motion
Stellar atmospheres	From cool to hot molecules or atomic plasmas. Specific tracers
Planetary atmospheres	From diffuse to dense, vast ranges
Extragalactic	Metallicities, star formation rates
Early universe	Only H, He e-; traces of D, Li

James Graham's Conception of The ISM as a Complex System



T and radiation compatible with chemistry

The Interstellar Medium
Astronomy 216
Spring 2008

Al Glassgold
University of California, Berkeley

Conditions of interstellar gases

- Large range in temperature & density
 $T \approx 10 - 10^6 \text{ K}$ $n \approx 10^{-3} - 10^6 \text{ cm}^{-3}$
- Even dense regions are “ultra-high vacuum”
 lab UHV: 10^{-10} Torr ($n \approx 4 \times 10^6 \text{ cm}^{-3}$)
 (STP conditions: $n = 3 \times 10^{19} \text{ cm}^{-3}$)

Ionized gases

H II regions - bright nebulae associated with regions of star formation & molecular clouds; ionized by stellar UV photons from early-type (OB) stars

$$T \approx 10^4 \text{ K} \quad \& \quad n_e \approx 0.1 - 10^4 \text{ cm}^{-3}$$

Warm Ionized Medium (WIM)

$$T \approx 8000 \text{ K} \quad \& \quad n_e \approx 0.025 \text{ cm}^{-3}$$

Hot Ionized Medium (HIM) - tenuous gas pervading the ISM, ionized by electron impact

$$T \approx 4.5 \times 10^5 \text{ K} \quad \& \quad n_e \approx 0.035 \text{ cm}^{-3}$$

Neutral gases

Cool clouds (CNM)

$$T \approx 100 \text{ K} \quad \& \quad n \approx 40 \text{ cm}^{-3}$$

Warm neutral gas (WNM)

$$T \approx 7500 \text{ K} \quad \& \quad n \approx 0.5 \text{ cm}^{-3}$$

Cold dark clouds ($M \approx 10 - 1000 M_{\odot}$)

$$T \geq 10 \text{ K} \quad \& \quad n \approx 10^2 - 10^4 \text{ cm}^{-3}$$

Giant molecular clouds ($M \approx 10^3 - 10^5 M_{\odot}$)

$$T \geq 20 \text{ K} \quad \& \quad n \approx 10^2 - 10^4 \text{ cm}^{-3}$$

with high density cores & clumps that form stars.

Where are molecules? (simplified)

Main gas depend on object scrutinized

- ISM, YSO, Stars :
H, H₂, (He), or cold plasma e⁻
- Comets : H₂O
- Atmospheres: H₂, CH₄, NH₃, N₂, SO₂, O₂
(Earth), mixtures

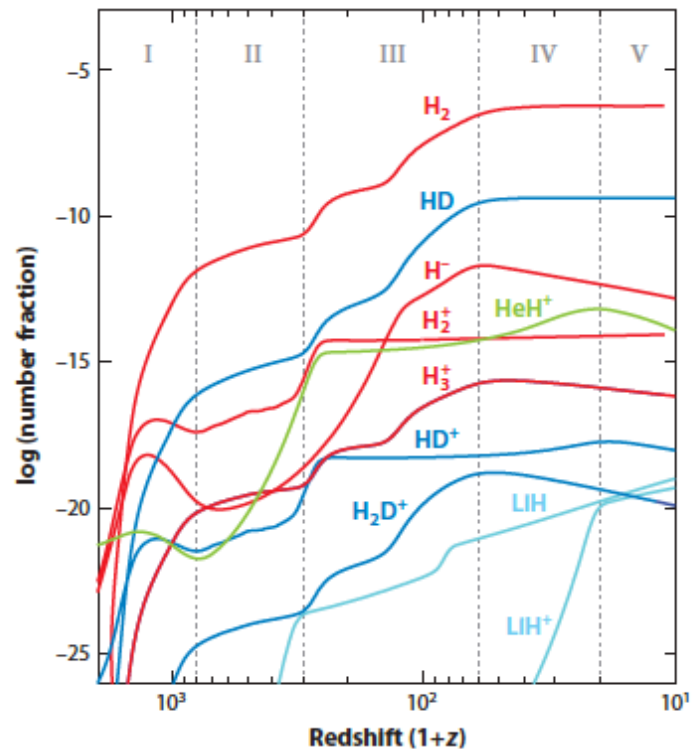
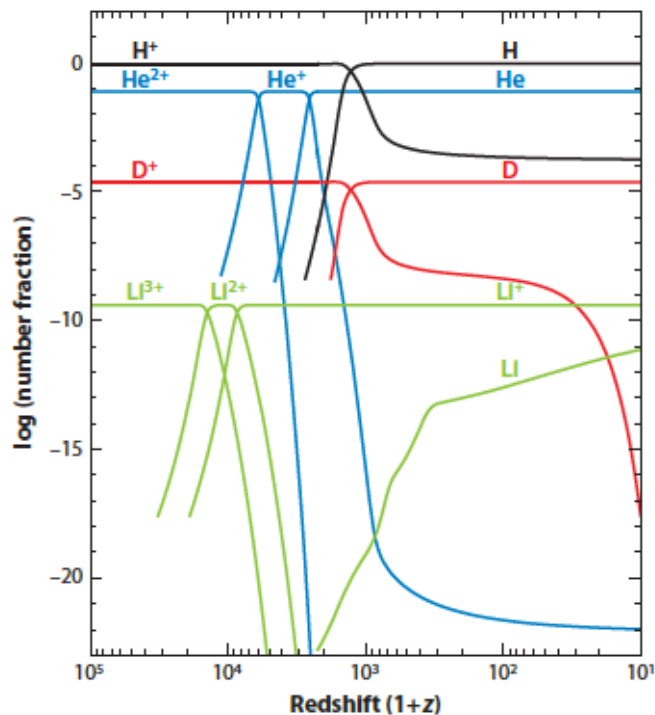
The main gas is very often optically inactive in microwave, not necessarily in IR
Active molecule as traces.

Special mention: early universe: Chemistry before stars

The Dawn of Chemistry

Daniele Galli and Francesco Palla

INAF-Osservatorio Astrofisico di Arcetri, Largo E. Fermi 5, 50125 Firenze, Italy;
email: galli@arcetri.astro.it, palla@arcetri.astro.it



Article

A Molecular Candle Where Few Molecules Shine: HeHHe⁺

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² Laboratoire Aimé-Cotton, CNRS & Université Paris-Saclay, 91405 Orsay, France; laurent.wiesenfeld@u-psud.fr

* Correspondence: r410@olemiss.edu; Tel.: +1-662-915-1687

† These authors contributed equally to this work

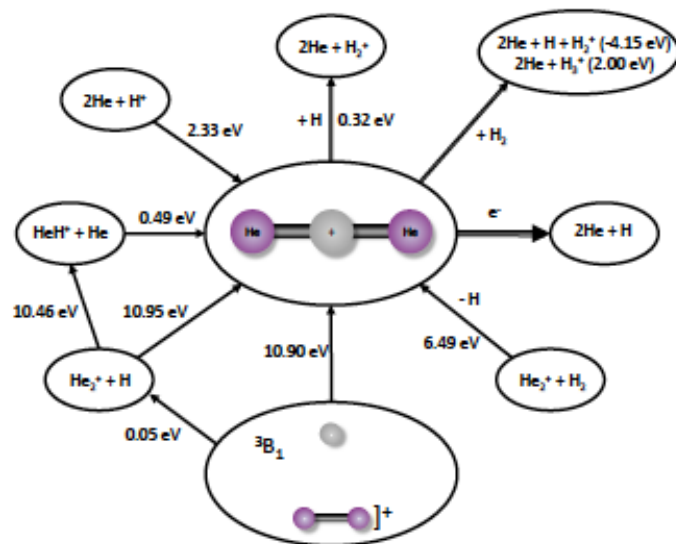


Figure 2. The pathways and ZPVE-corrected energetics for the creation and destruction of ${}^1\Sigma_g^+$ HeHHe⁺. Positive energies favor the products based on the direction of the arrow. Negative energies favor the reactants.

First astrophysical detection of the helium hydride ion (HeH⁺)

 Rolf Güsten¹, Helmut Wiesemeyer¹, David Neufeld², Karl M. Menten¹, Urs U. Graf³, Karl Jacobs³, Bernd Klein^{1,4}, Oliver Ricken¹, Christophe Risacher^{1,5}, Jürgen Stutzki³

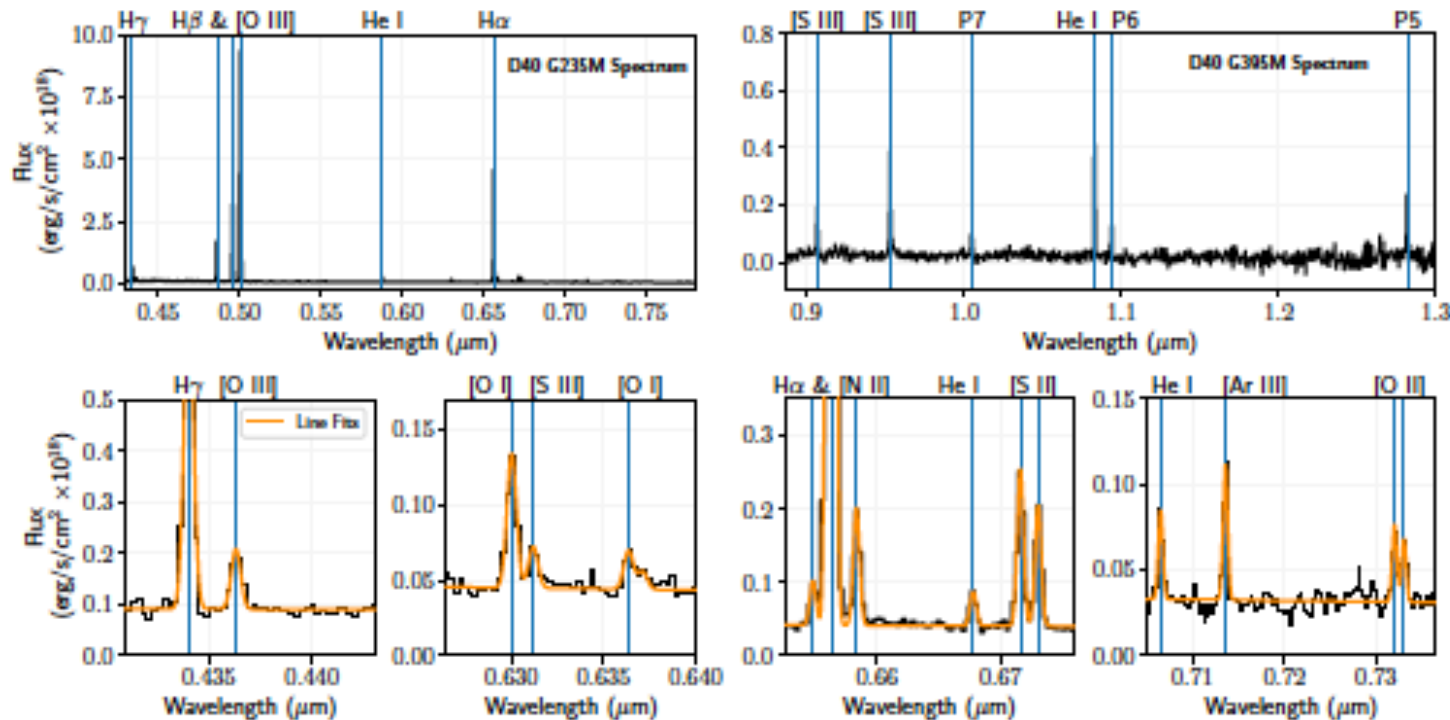
Extra galactic

CECILIA: Direct O, N, S, and Ar Abundances in Q2343-D40, a Galaxy at $z \sim 3$

NOAH S. J. ROGERS,¹ ALLISON L. STROM,^{1,2} GWEN C. RUDIE,³ RYAN F. TRAINOR,⁴ MENELAOS RAPTIS,⁴ AND CAROLINE VON RAESFELD^{1,2}

DIRECT GAS-PHASE ABUNDANCES AT $z \sim 3$

3



IMAGES, SPECIFIC CASES



* **Dense clouds**

$T \sim 5-10 \text{ K}$, $n \sim 10^5 \text{ cm}^{-3}$

$x_e \sim 10^{-8}$

Tracers : CO, HCO+, N₂H+, H₂D+

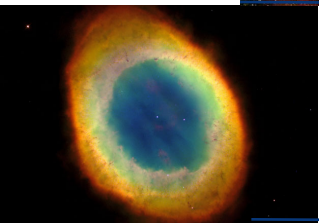


* **Diffuse clouds**

$T \sim 100 \text{ K}$, $n \sim 10^2 \text{ cm}^{-3}$

$x_e \sim 10^{-4}$

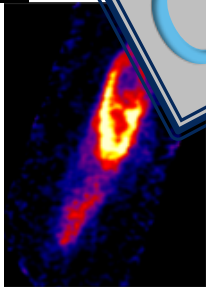
Tracers : OH, CH⁺, HCO⁺



* **Photodissociation regions**

$T \sim 1000 \text{ K}$

$n \sim 10^4 \text{ cm}^{-3}$
 $x_e \sim 10^{-5}$
 Tracers : C⁺, Si⁺, S⁺



* **Jets**

$T \sim 10-1000 \text{ K}$, $n \sim 10^{4-7} \text{ cm}^{-3}$

Tracers : CO, HCO+, SiO



**Star forming regions/
YSO environments**

$T \sim 5-300 \text{ K}$, $n \sim 10^{4-7} \text{ cm}^{-3}$

$x_e \sim 10^{-9-1a}$

Tracers : CO, HCO+, H₂O

H₂CO, CH₃OH,

N, S containing molecules

Oldies but Goodies



*** Dense clouds**

$T \sim 5-10 \text{ K}$, $n \sim 10^5 \text{ cm}^{-3}$

$x_e \sim 10^{-8}$

Tracers : CO, HCO⁺, N₂H⁺, H₂D⁺

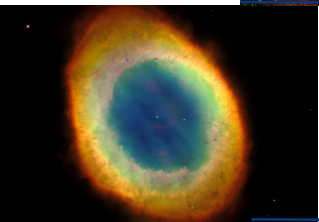


*** Diffuse clouds**

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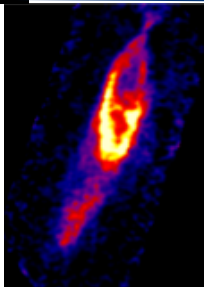


*** Photon-Dominated Regions**

$T \sim 1000 \text{ K}$, $n \sim 10^4 \text{ cm}^{-3}$

$x_e \sim 10^{-4}$

Tracers : CN, CH, CH⁺, C⁺



*** Jets**

$T \sim 10-1000 \text{ K}$, $n \sim 10^{4-7} \text{ cm}^{-3}$

Tracers : CO, HCO⁺, SiO



**Star forming regions/
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$T \sim 5-300 \text{ K}$, $n \sim 10^{4-7} \text{ cm}^{-3}$

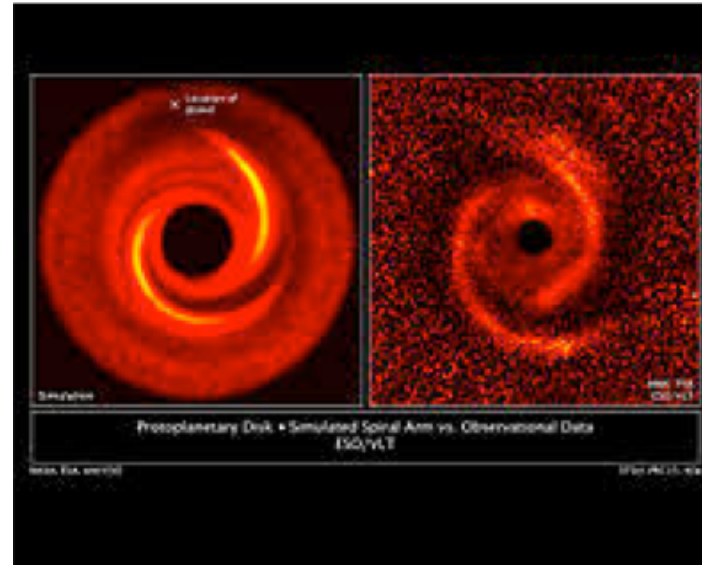
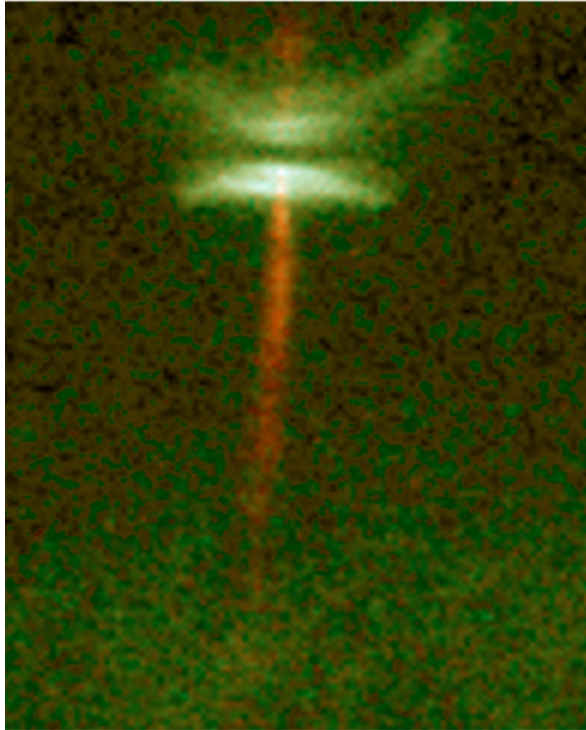
$x_e \sim 10^{-9-1a}$

Tracers : CO, HCO⁺, H₂O

H₂CO, CH₃OH,

N, S containing molecules

Protoplanetary disks



Protoplanetary Disks

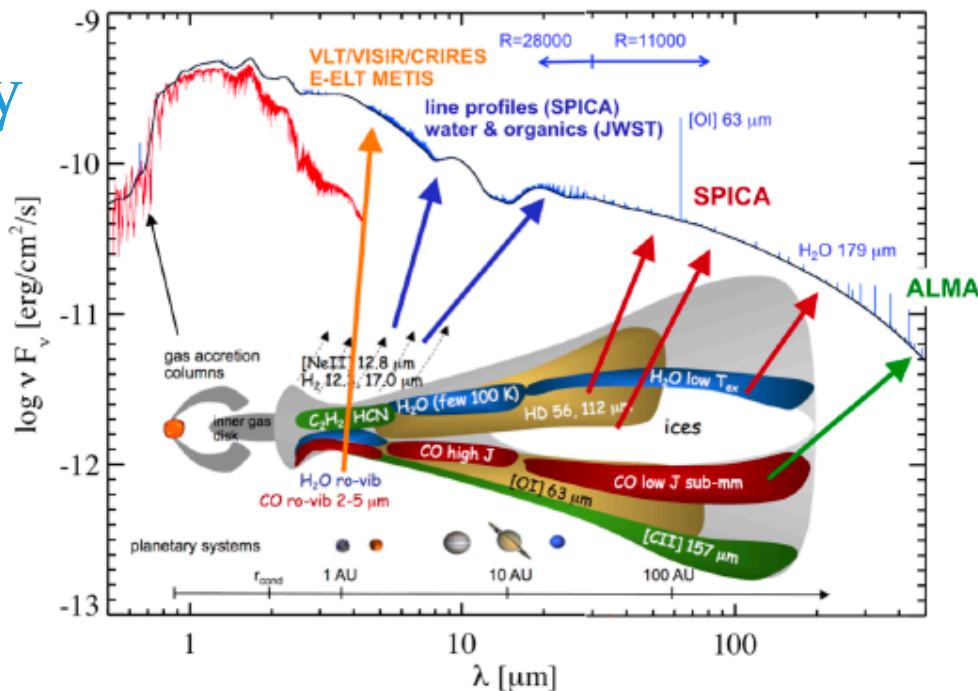


Figure 2. An overview of line emission from near-IR to submm wavelengths from planet forming disks and existing and upcoming instrumentation to detect it. The model SED is that of a typical T Tauri disk using two different spectral resolutions ($R = 28\,000$ and $R = 11\,000$ relevant for the planned SMI and SAFARI instruments onboard the proposed SPICA mission).

Postpone the atmospheres...

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$1 \text{ kcal/mole} \approx 43 \text{ meV} \approx 503 \text{ K} \approx 350 \text{ cm}^{-1} \approx 10 \text{ THz}$

Chemistry

Thermodynamics

Spectroscopy

Electron, atomic, nuclear
physics

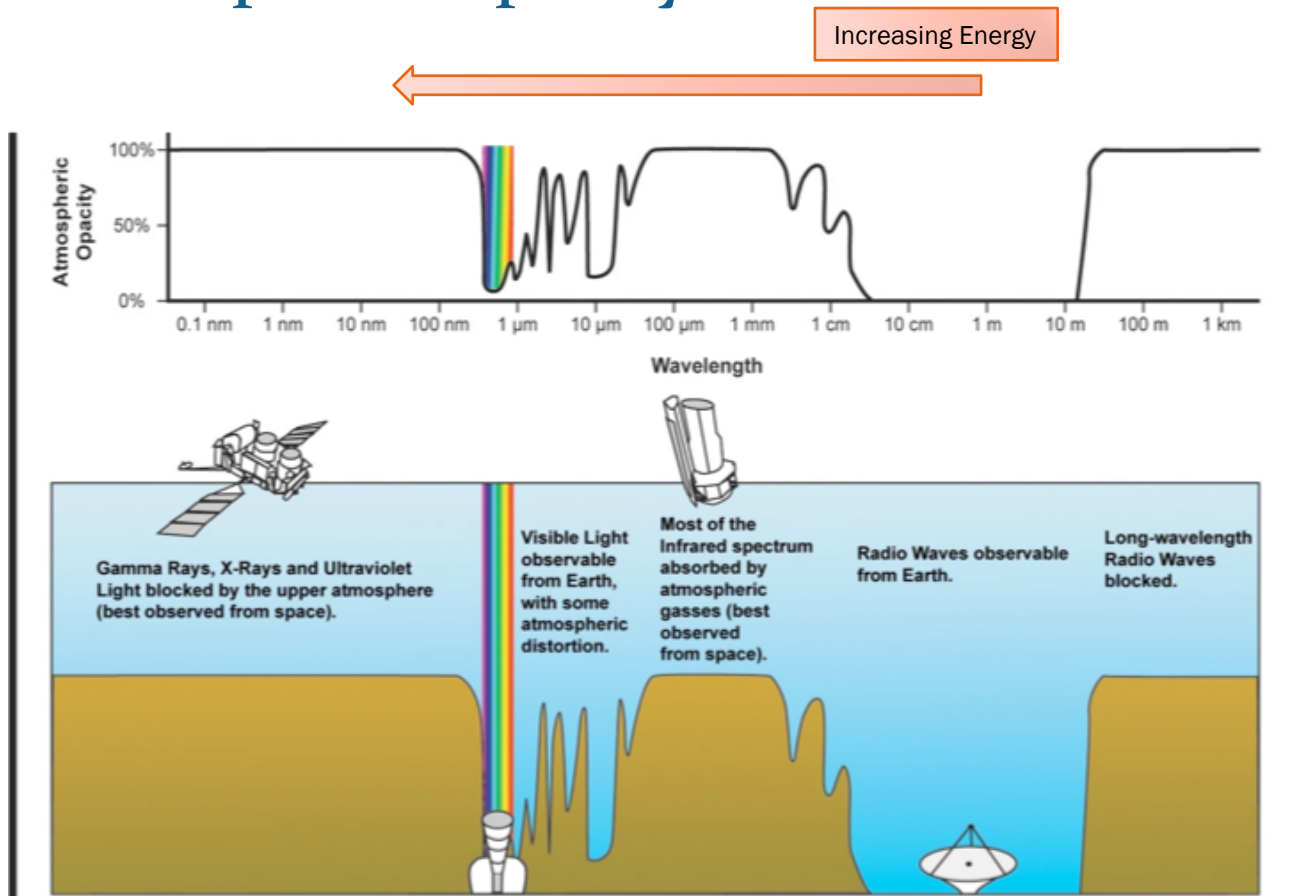
How to detect? A note on units

- **Probes:** many ways to deal with molecules, from chromatography to mass spectrometry. Valid for example in our atmosphere, for planetary probes (Cassini, Rosetta,...)
- **Telescopes:** wavelengths from cm to sub-Å, from radio waves to gamma-rays, from μeV to GeV (cosmic rays). Single optics (dish or mirror) or interferometry (ALMA, Plateau de Bures, VLA, VLTI, Gemini)

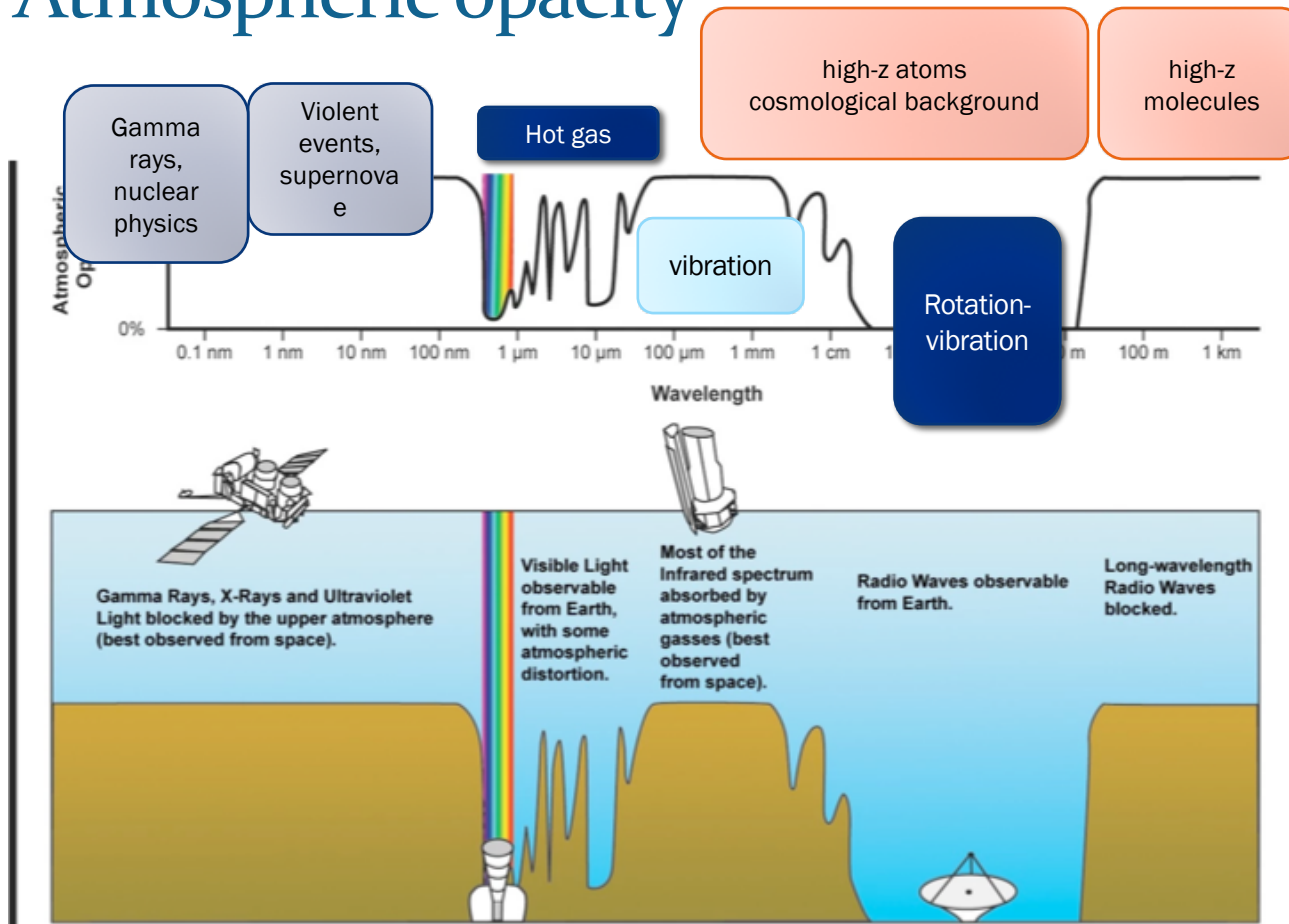
Quantum states		Energies	Type of radiation	How to detect
Hyperfine structure		kHz → MHz	Longwave to radio	Usually indirectly detected
Fine structure (mostly atoms)		100GHz → a few THz	microwave to FIR	ground & space Favorable for high-z atomic gasses
Rotation		0.5 → 60 cm ⁻¹	millimeter to sub-mm	ground (mostly) and space / plane/balloons
Vibration		100 → A few 1000 cm ⁻¹	FIR to IR	Space and balloons Favorable for high-z molecular gasses
Electronic transitions	Molecules	up to a few eV	visible, UV	ground and space
	Atoms	up to a few keV	Visible → X-rays	ground and space

$$1 \text{ kcal/mole} \approx 43 \text{ meV} \approx 503 \text{ K} \approx 350 \text{ cm}^{-1} \approx 10 \text{ THz}$$

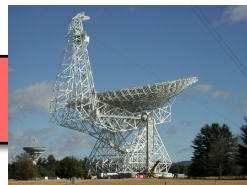
Atmospheric opacity



Atmospheric opacity



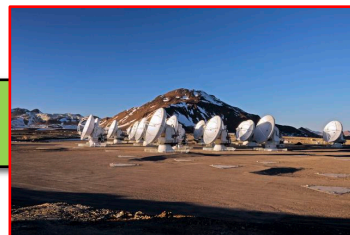
GBT, VLA < 50 GHz



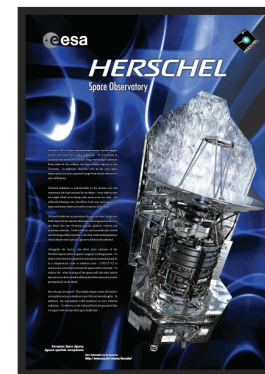
IRAM 80... 270 GHz



ALMA 80... 950 GHz



Herschel 480 ... 1910 GHz (HIFI)



Quantum states		Energies	Type of radiation	How to detect
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Young stellar object: solar mass binary

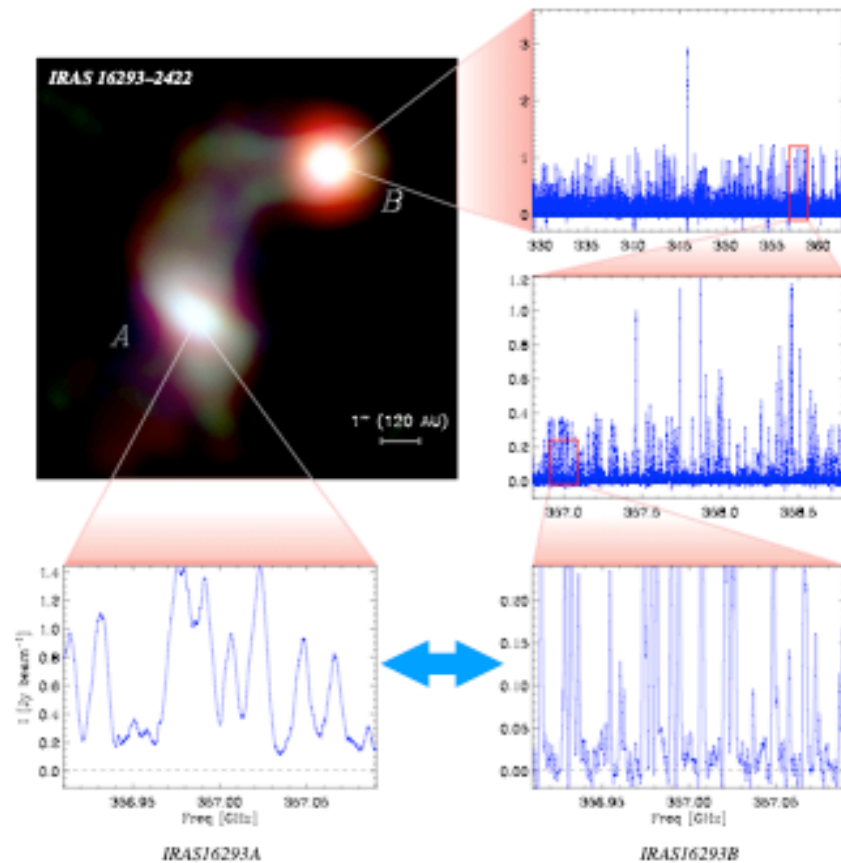
The ALMA-PILS survey: inventory of complex organic molecules towards IRAS 16293-2422 A

J. Jørgensen et al., Niels Bohr Inst. et al.

Table 1 List of molecules mentioned in this review with more than three atoms.

Species	Formula	Species	Formula
Acetaldehyde	CH ₃ CHO	Glyoxal	HC(O)CHO
Acetamide	CH ₃ C(O)NH ₂	Hydroxylamine	NH ₂ OH
Acetic acid	CH ₃ COOH	Isocyanic acid	HNCO
Acetone	CH ₃ C(O)CH ₃	Methane	CH ₄
Ammonia	NH ₃	Methanimine	CH ₂ NH
Benzene	c-C ₆ H ₆	Methanol	CH ₃ OH
Benzonitrile	c-C ₆ H ₅ CN	Methoxymethanol	CH ₃ OCH ₂ OH
Butyl cyanide	C ₄ H ₉ CN	Methyl acetylene	CH ₃ CCH
Cyanoacetylene	HC ₃ N	Methyl amine	CH ₃ NH ₂
Cyanodiacetylene	HC ₅ N	Methyl chloride	CH ₃ Cl
Cyanoformaldehyde	NCCHO	Methyl cyanide	CH ₃ CN
Cyanomethanimine	NHCHCN	Methyl formate	CH ₃ OCHO
Cyanomethyl radical	CH ₂ CN	Nitrous acid	HONO
Cyclopropenone	c-H ₂ C ₃ O	Propanal	C ₂ H ₅ CHO
Dimethyl ether	CH ₃ OCH ₃	Propanol	C ₃ H ₇ OH
Ethanamine	CH ₃ CHNH	Propenal	C ₂ H ₃ CHO
Ethanol	C ₂ H ₅ OH	Propyl cyanide	C ₃ H ₇ CN
Ethylene glycol	(CH ₂ OH) ₂	Propylene oxide	c-CH(CH ₃)CH ₂ O
Formaldehyde	H ₂ CO	Quinoline	C ₉ H ₇ N
Formamide	NH ₂ CHO	Thioformaldehyde	H ₂ CS
Formic acid	HCOOH	Vinyl cyanide	C ₂ H ₃ CN
Glycolaldehyde	CH ₂ (OH)CHO	Urea	NH ₂ C(O)NH ₂
Glycolonitrile	HOCH ₂ CN		

The ALMA-PILS survey: inventory of complex organic molecules towards IRAS 16293-2422 A



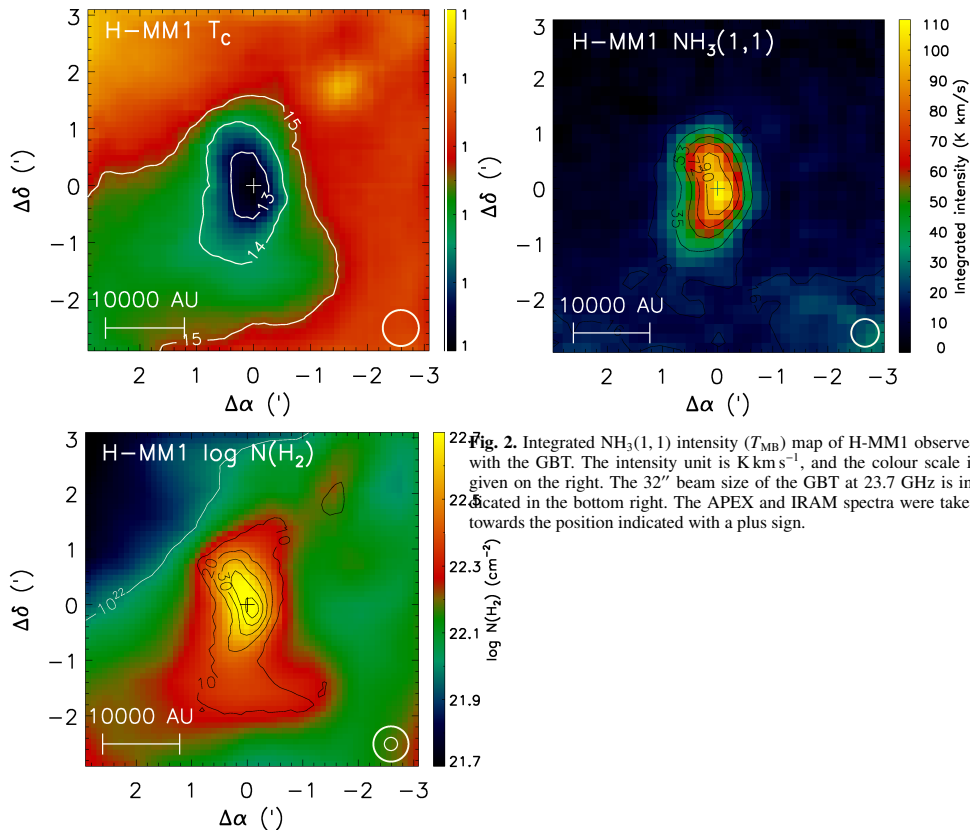


Fig. 1. Dust colour temperature (T_c , top) and the H_2 column density ($N(\text{H}_2)$, bottom) maps of H-MM1 as derived from Herschel/SPIRE maps at 250, 350, and 500 μm . The distribution of the 850 μm emission observed with SCUBA-2 is indicated with black contours on the $N(\text{H}_2)$ map. The contour levels are 10 to 50 by 10 MJy sr^{-1} . The column

Fig. 2. Integrated $\text{NH}_3(1,1)$ intensity (T_{MB}) map of H-MM1 observed with the GBT. The intensity unit is K km s^{-1} , and the colour scale is given on the right. The 32'' beam size of the GBT at 23.7 GHz is indicated in the bottom right. The APEX and IRAM spectra were taken towards the position indicated with a plus sign.

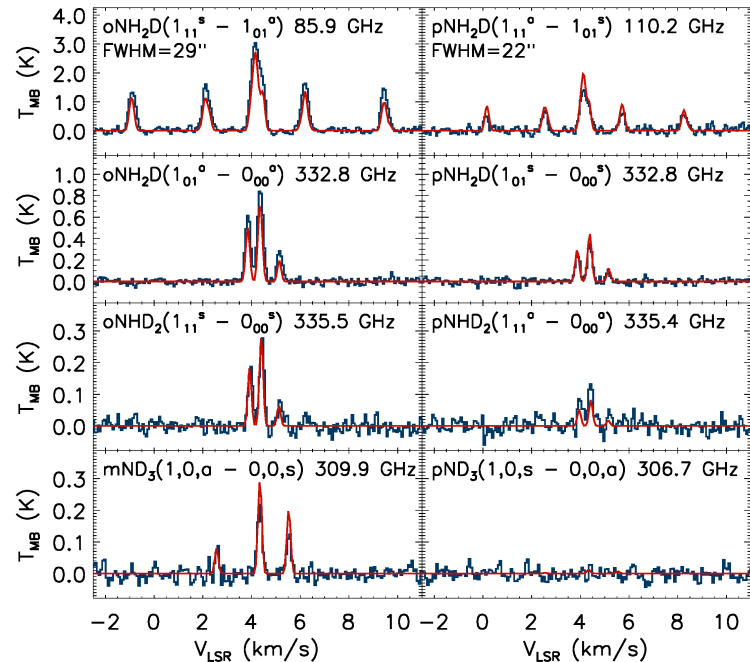


Fig. 16. Deuterated ammonia spectra produced by the core model at the time 3×10^5 yr (red curves) together with the observed spectra (histograms).

Starless core Ophiuchus/H-MM1

Harju et al. 2016

Stellar atmosphere : case of brown dwarf

A 1.46–2.48 μm spectroscopic atlas of a T6 dwarf (1060 K) atmosphere with IGRINS: first detections of H_2S and H_2 , and verification of H_2O , CH_4 , and NH_3 line lists

Tannock et al., MNRAS, 2022

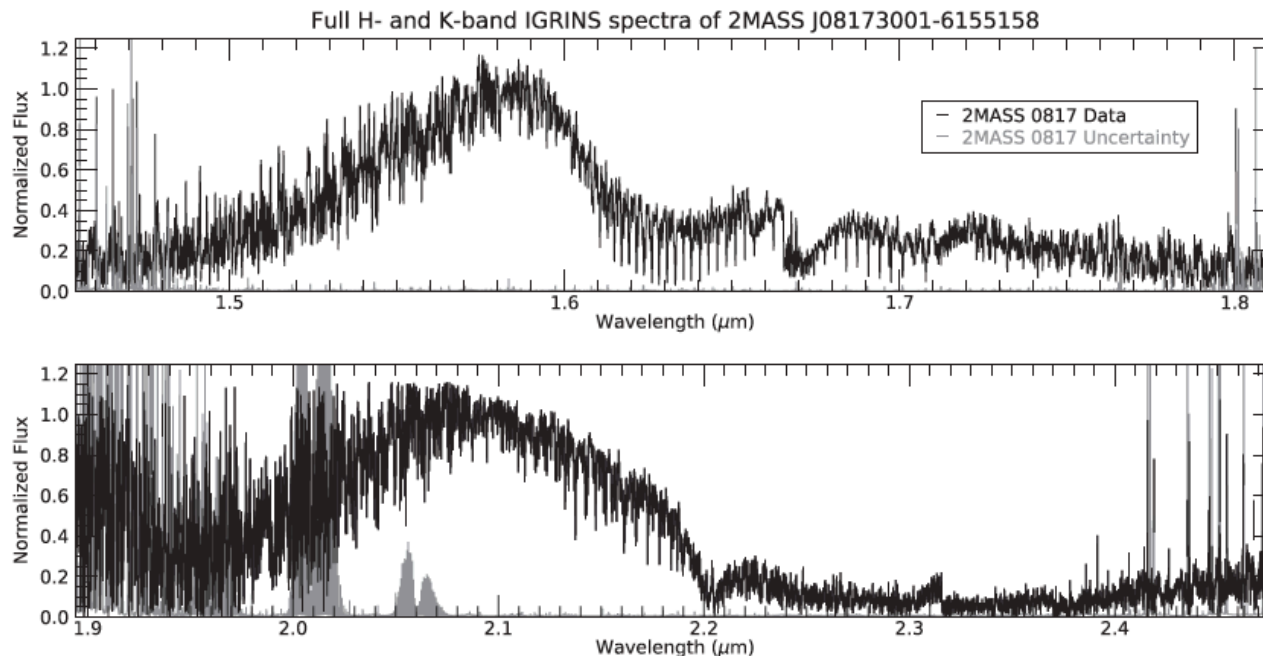


Figure 2. The full *H*- and *K*-band IGRINS spectra of 2MASS J08173001–6155158 with epochs combined and the orders stitched together. This figure does not include the quadratic correction described in Section 3.1. These data appear noisy, but in fact have $S/N \simeq 300$ at the peak of the *H*-band spectrum and $S/N \simeq 200$ at the *K*-band peak. The apparent noise spikes are all absorption features, and can be seen in detail in the full set of figures in the Appendix, available in the Online Supplementary Material.

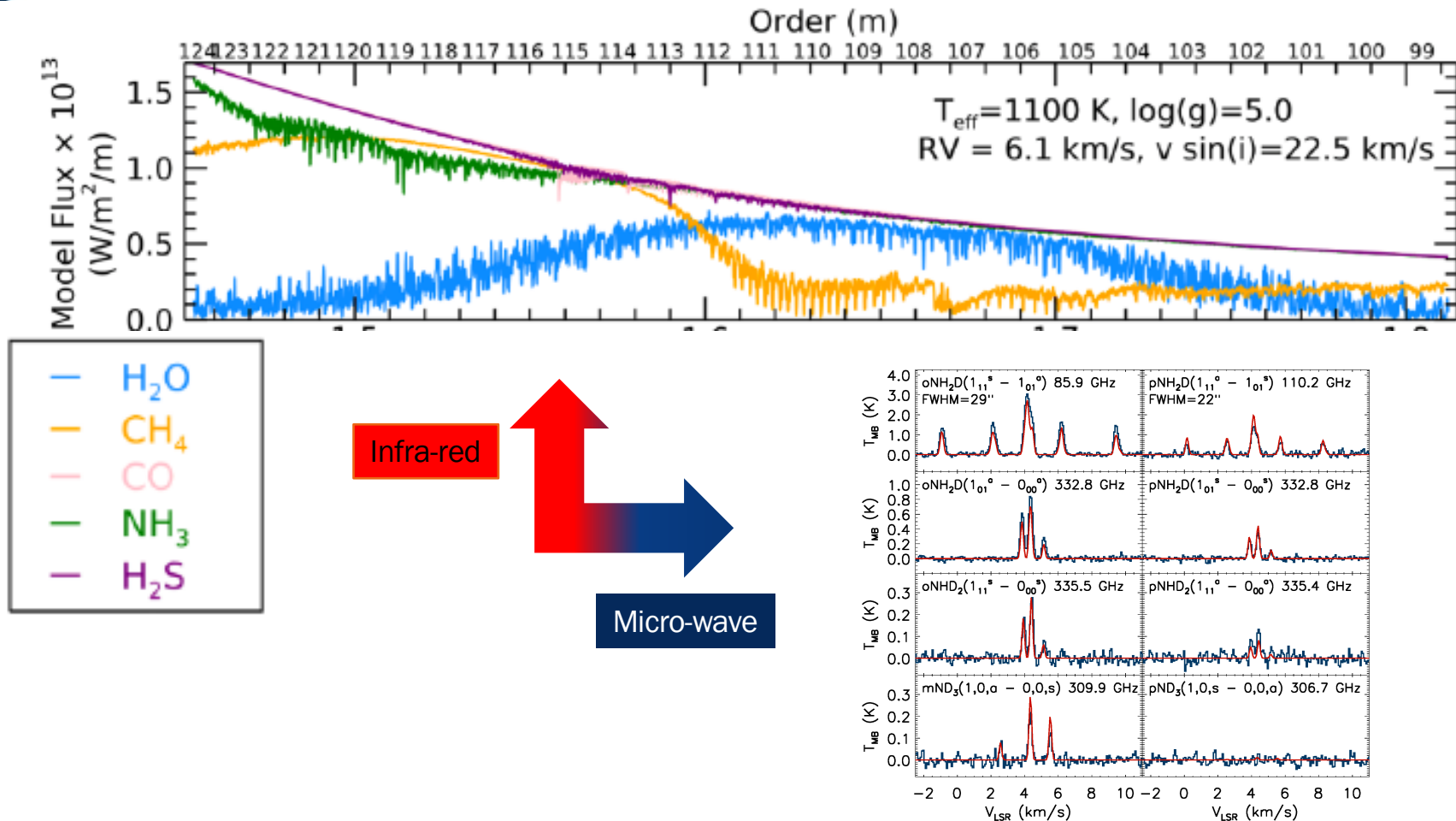
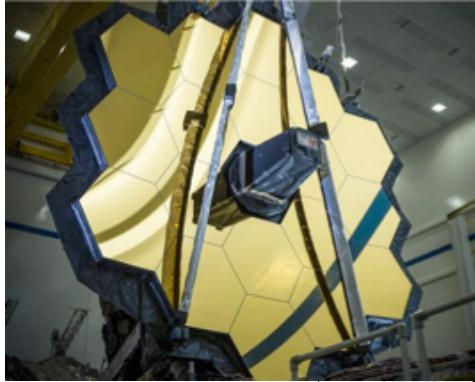
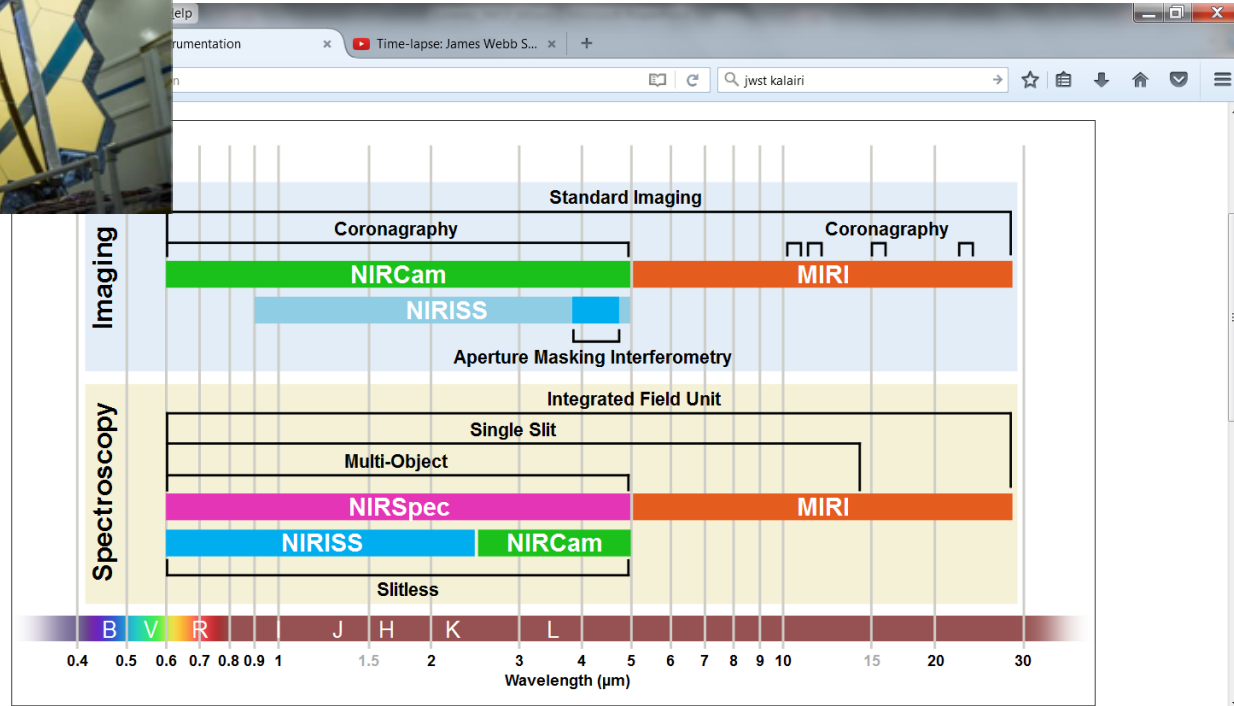


Fig. 16. Deuterated ammonia spectra produced by the core model at the size 2.10^5 au (red curves) together with the observed spectra (blue curves).



Special topic : JWST



Protoplanetary Disks

THE ASTROPHYSICAL JOURNAL LETTERS, 957:L22 (14pp), 2023 November 10

Banzatti et al.

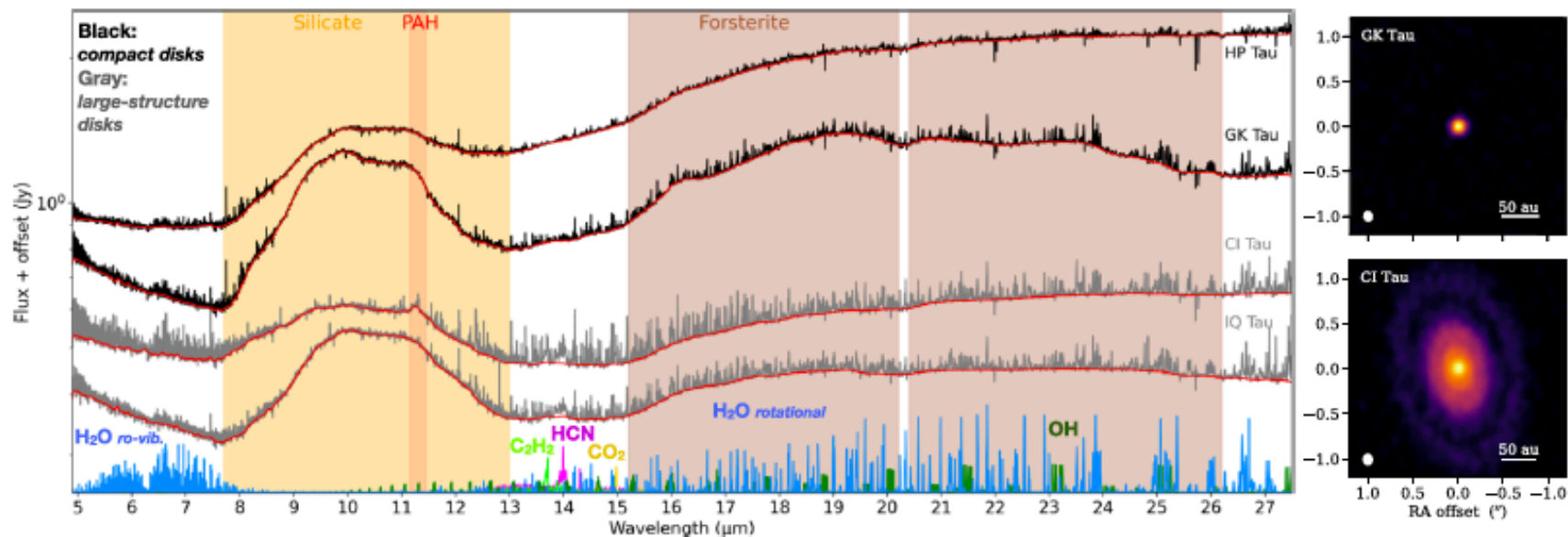
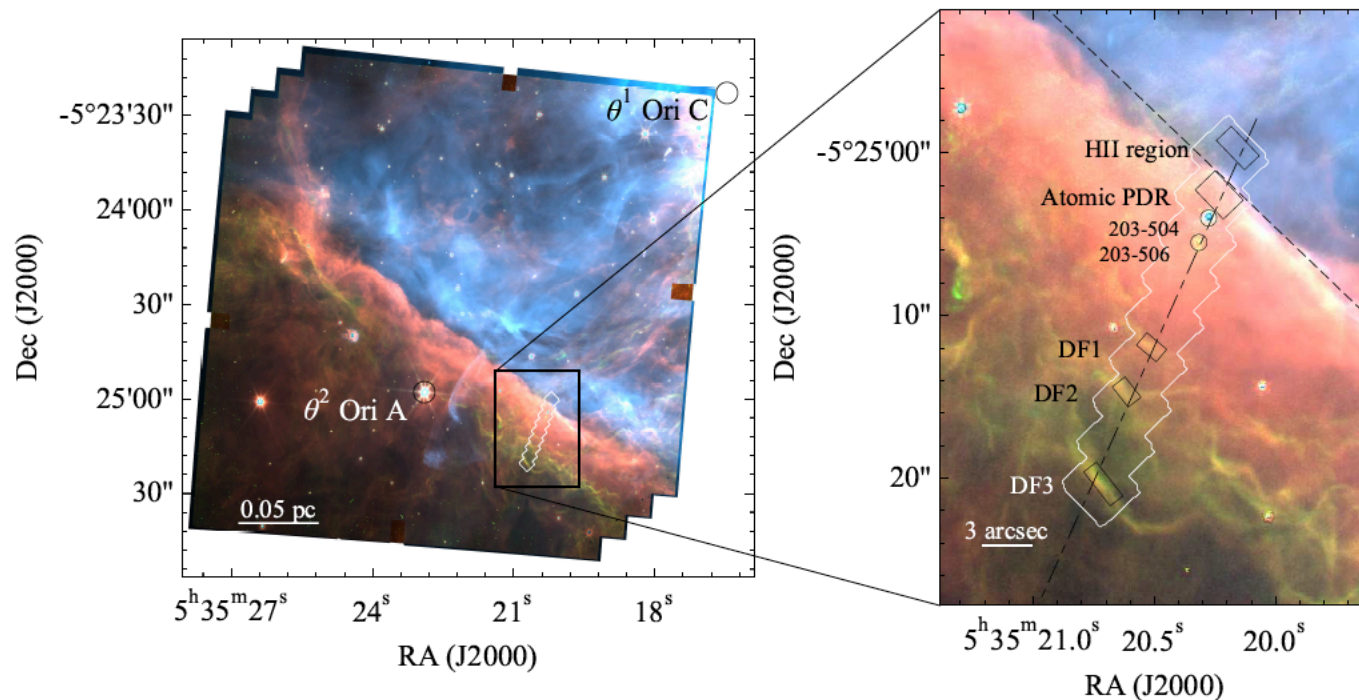


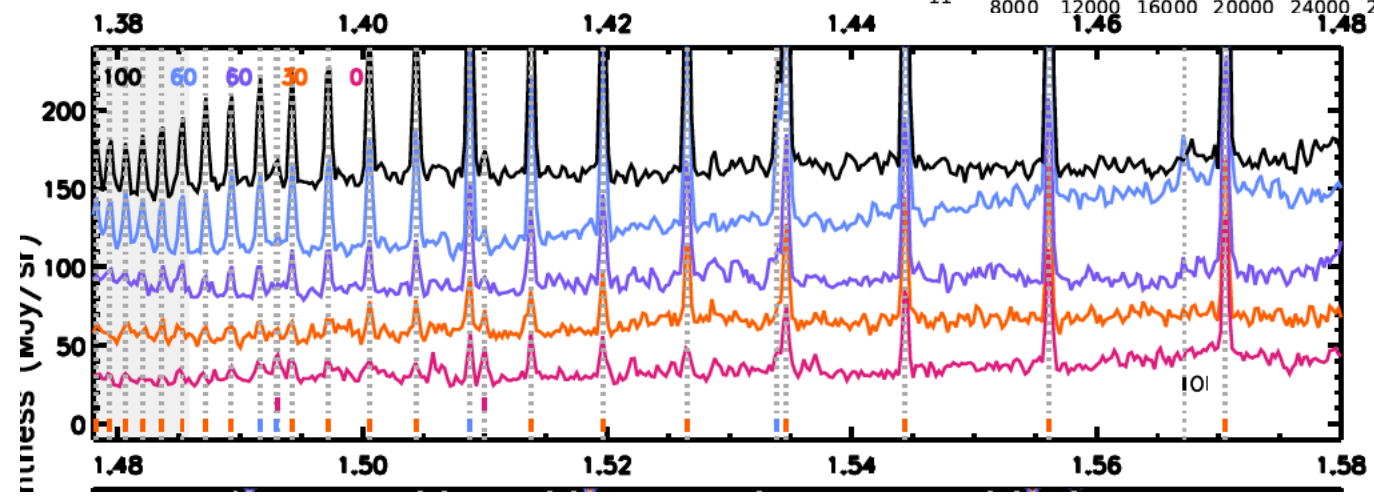
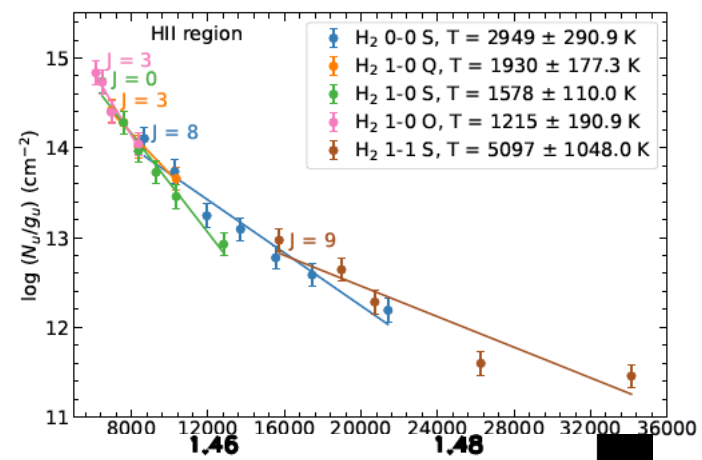
Figure 2. JWST-MIRI-MRS spectra for the four disks. For clarity, the spectra are offset vertically by the following additive shifts: 0.05 Jy in CI Tau, 0.17 Jy in GK Tau, and 0.4 Jy in HP Tau. Illustrative models of molecular emission are shown for guidance at the bottom. Prominent dust features are approximately identified and marked with shaded regions. The estimated continuum that is subtracted before analyzing the water spectra is shown in red on each spectrum. Two ALMA images are shown on the right for reference regarding disk sizes and structures (the whole sample is included in Appendix A).

PDR (photon dominated region): here the Orion Bar

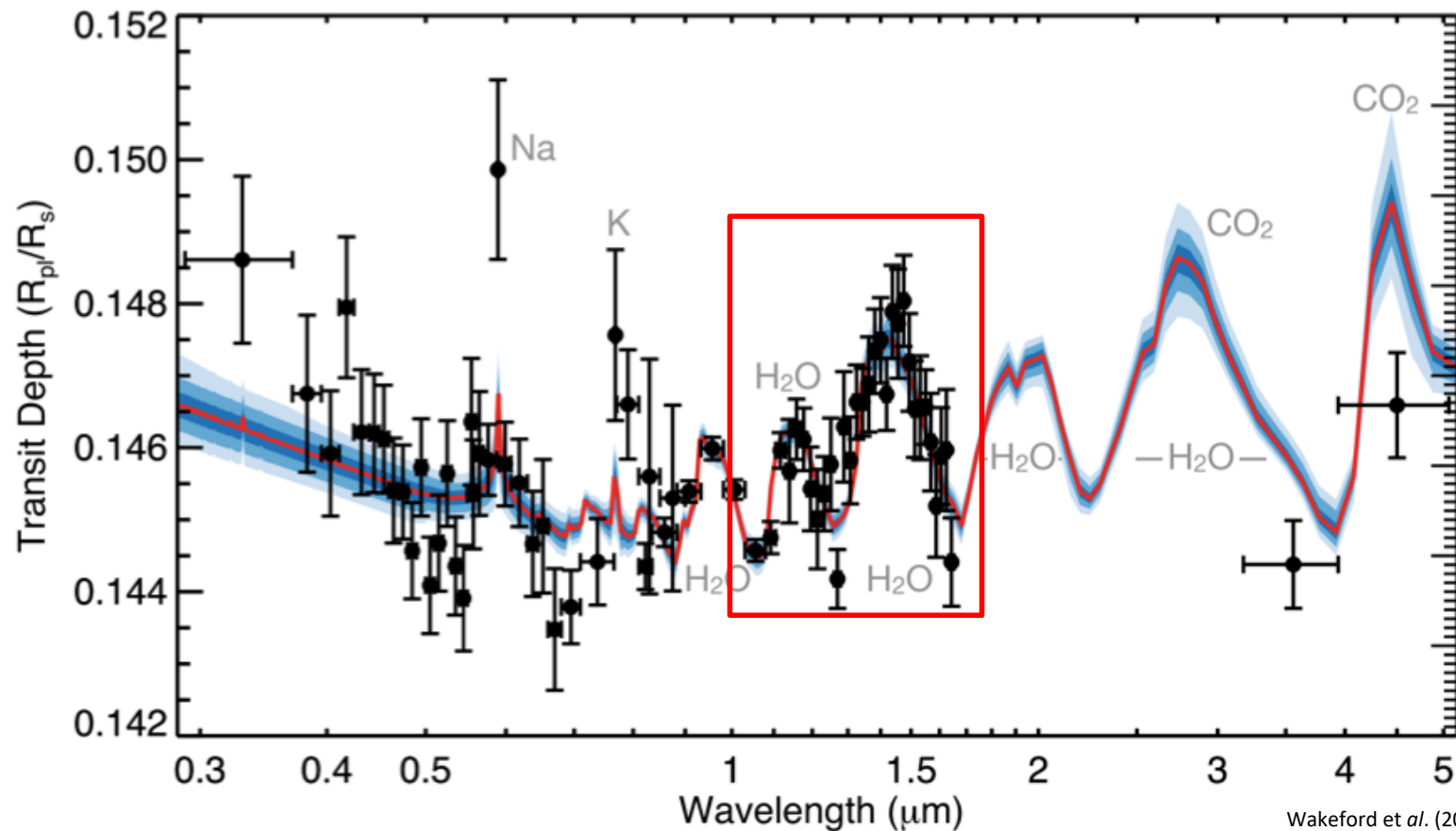


H₂ $\Delta j=0, \pm 2$ detection

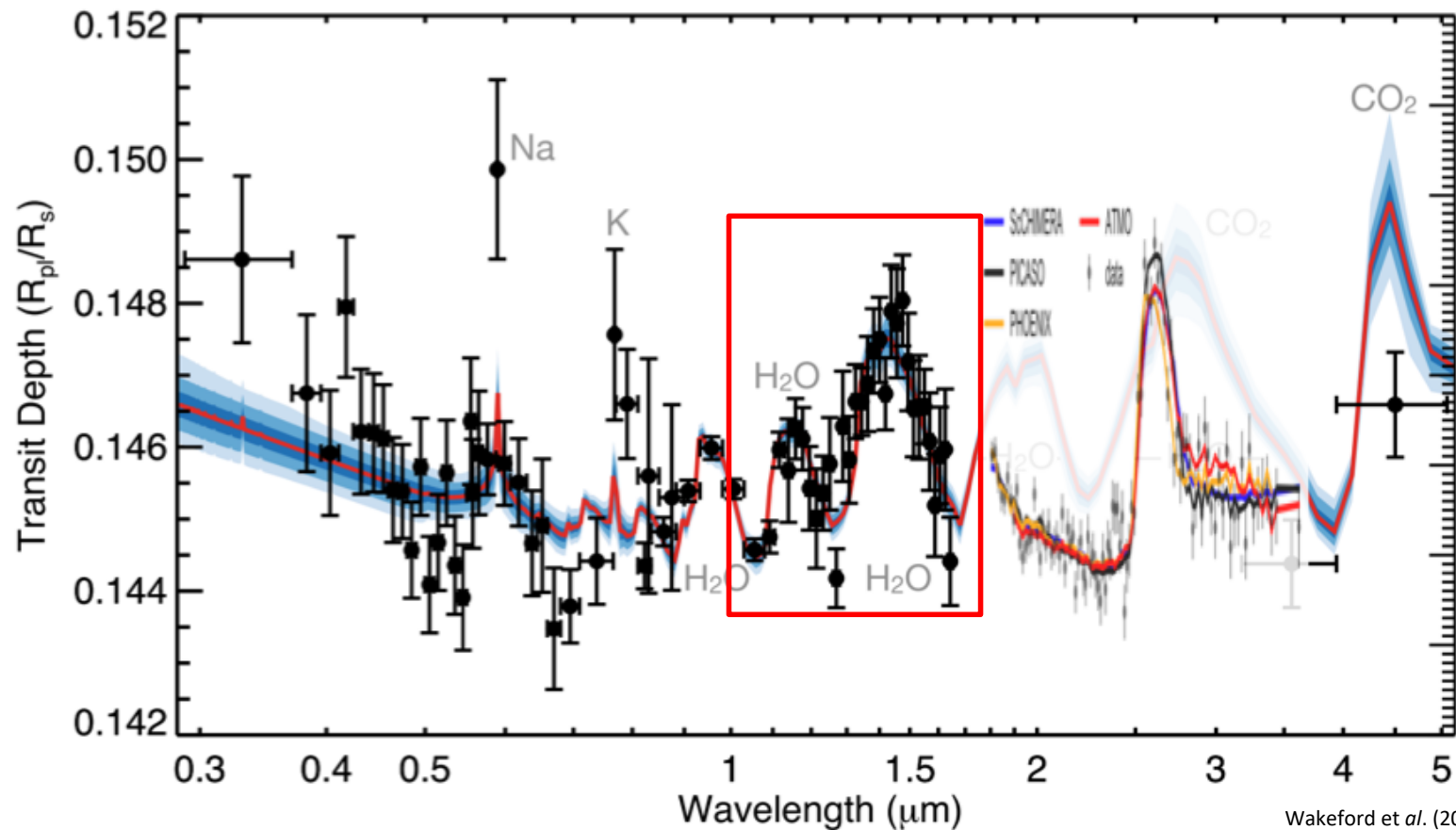
Quadrupolar (E₂) transitions



Before and after JWST



Before and after JWST



Table

1. Why molecules?

- It all begins with nuclei
- Relevance

2. Where molecules?

- Hot to cold
- Dense to thin
- Ionized, Plasmas, PDRs

3. When molecules?

- Early universe
- Disks
- Planets

4. How to detect?

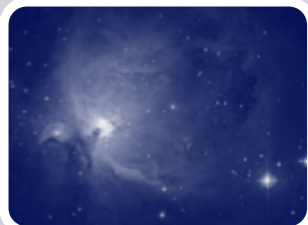
- Rotation
- IR and other

5. How to model?

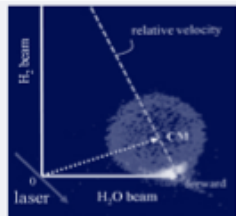
- Qualitative/Quantitative
- Ab initio

6. Conclusion

The tripod

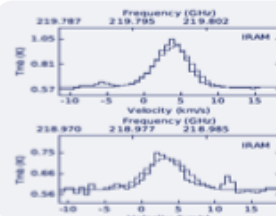


Observations

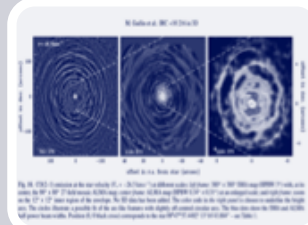


Laboratory

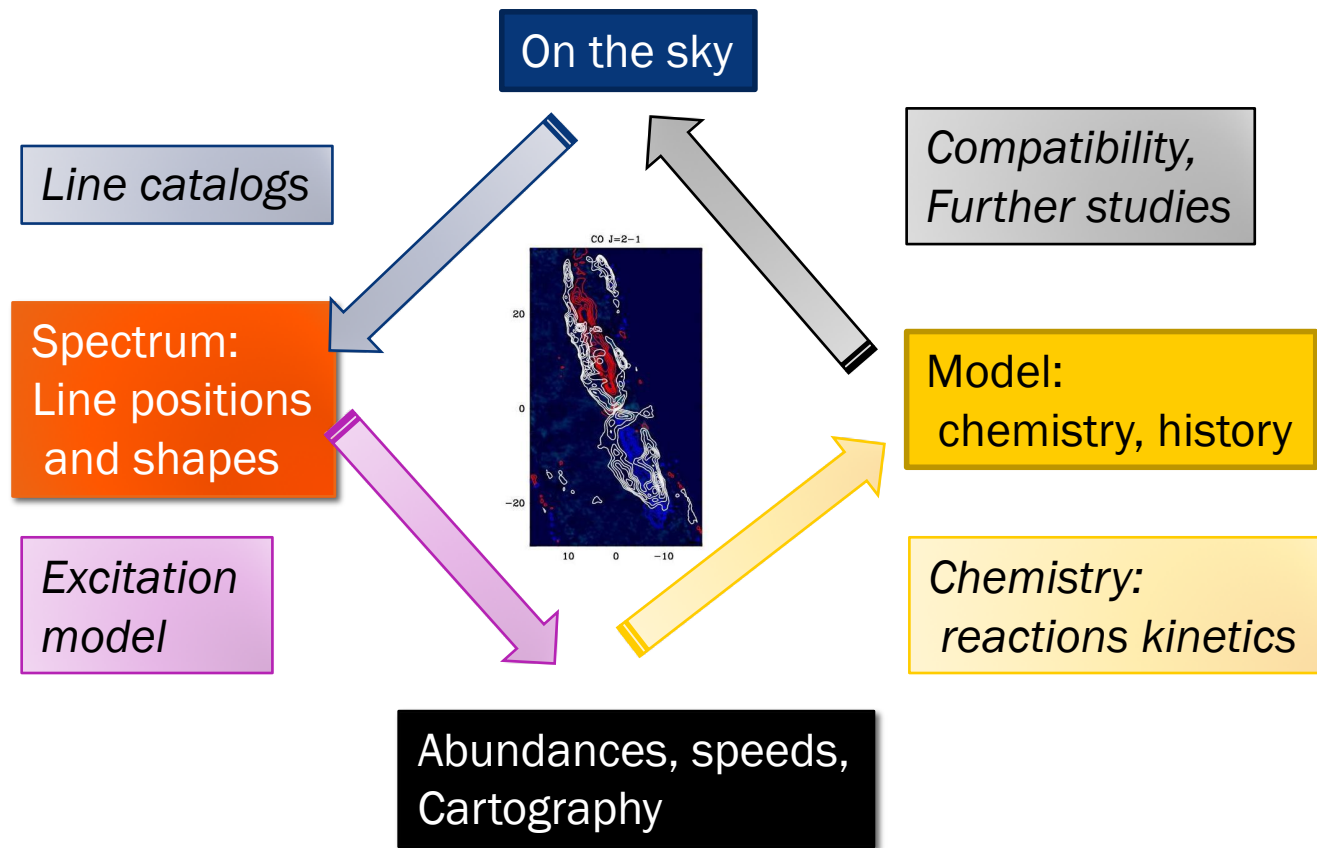
- Experiments
- Theory



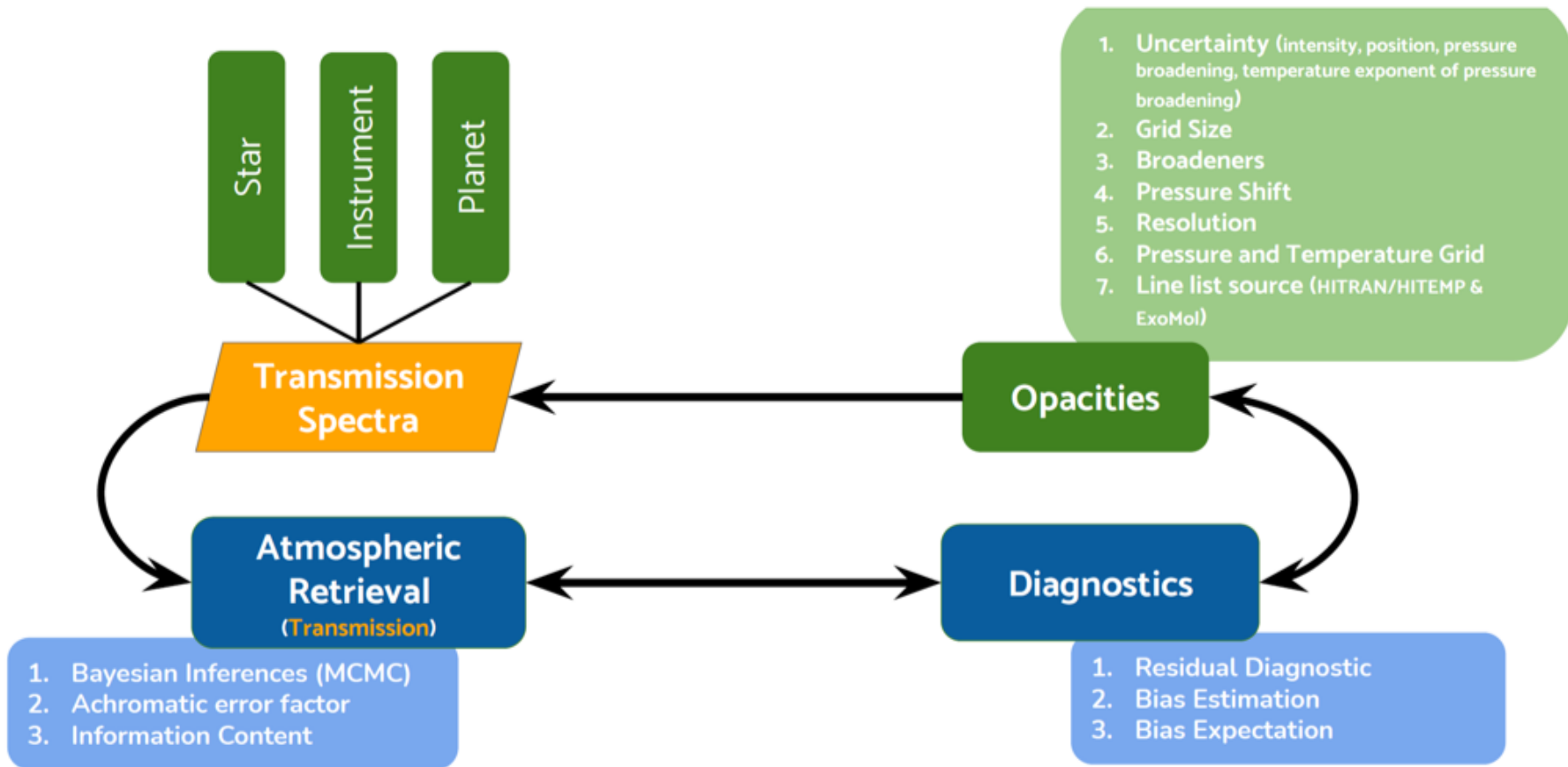
Models



Object



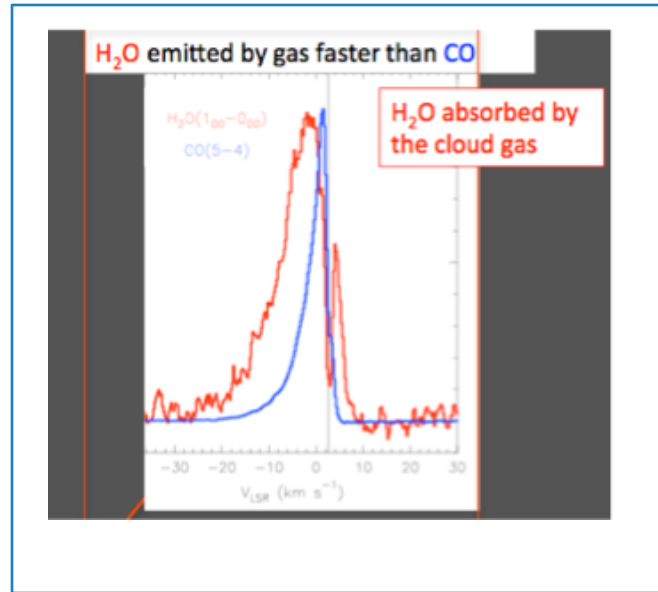
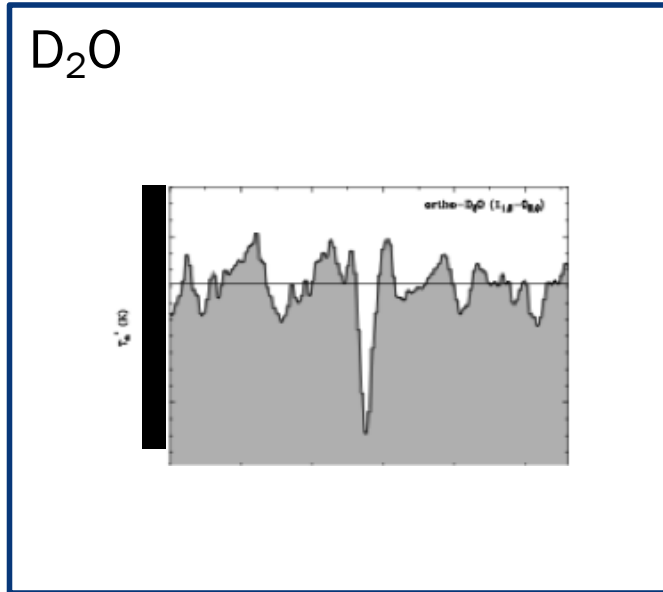
Assessing the extent of the opacity challenge for exoplanetary sciences



MICROSCOPIC DATA, COLLISIONS

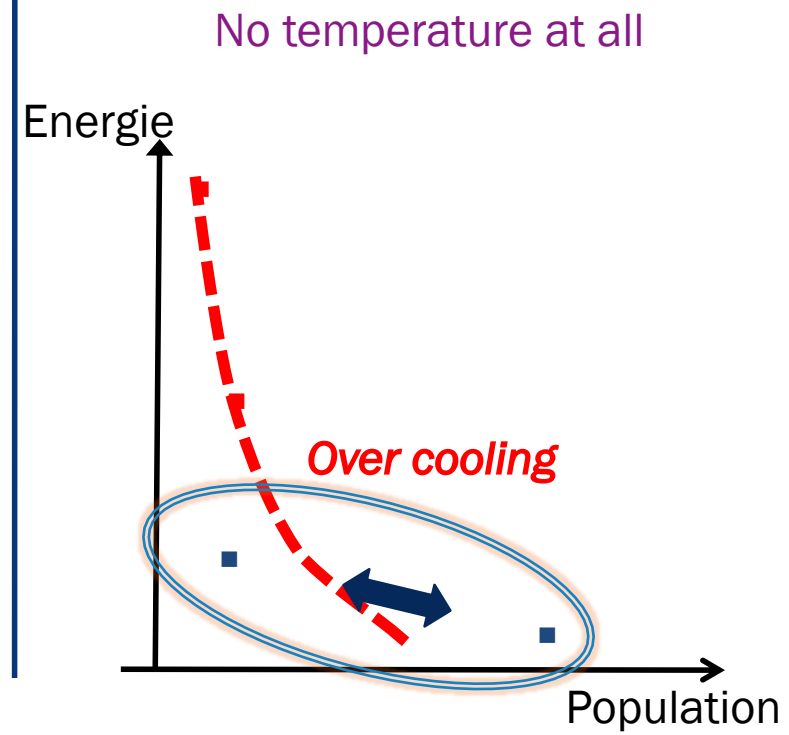
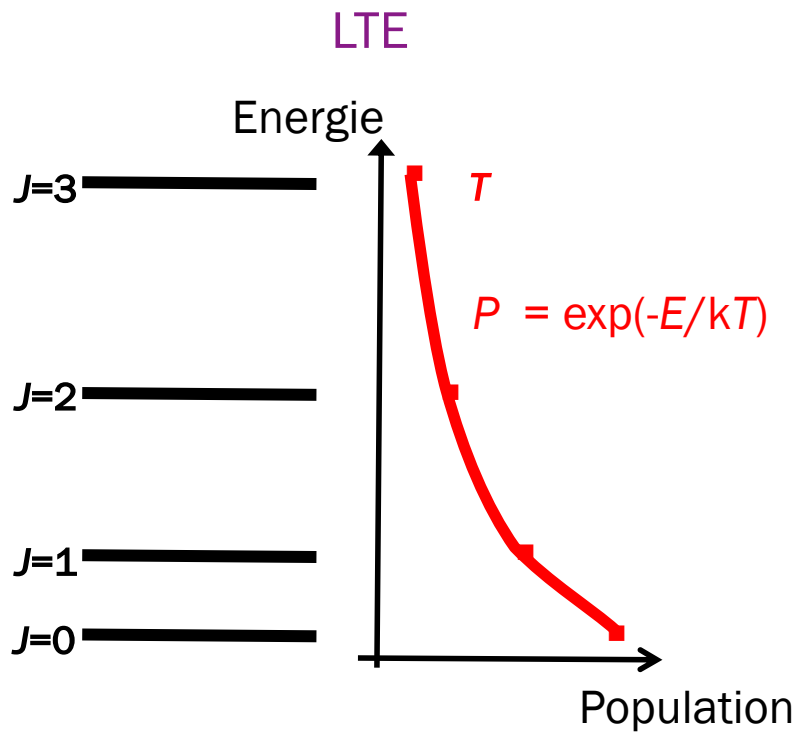
Our work in Grenoble, than Paris

Rotational spectrum detected by HIFI /Herschel

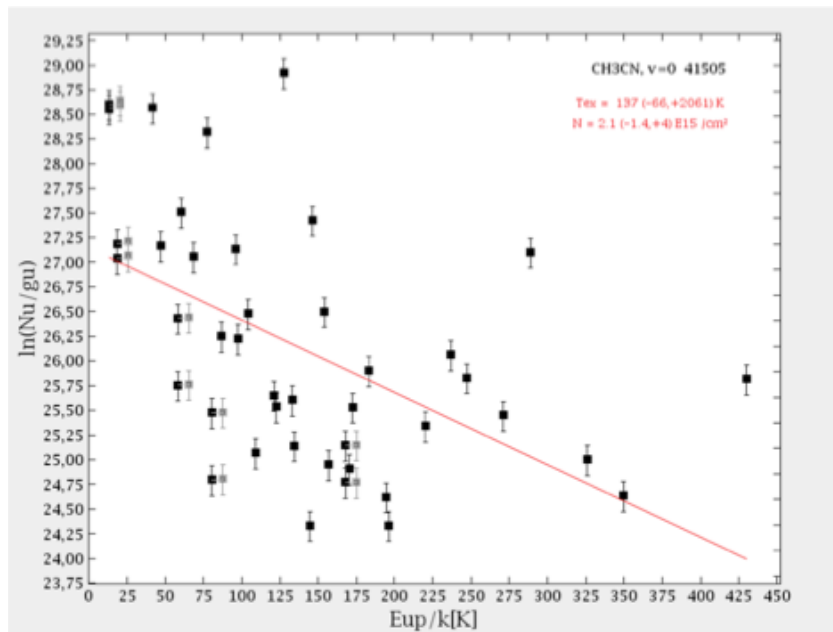


(the CHESS collaboration, Ceccarelli et al., star forming regions and shocks)

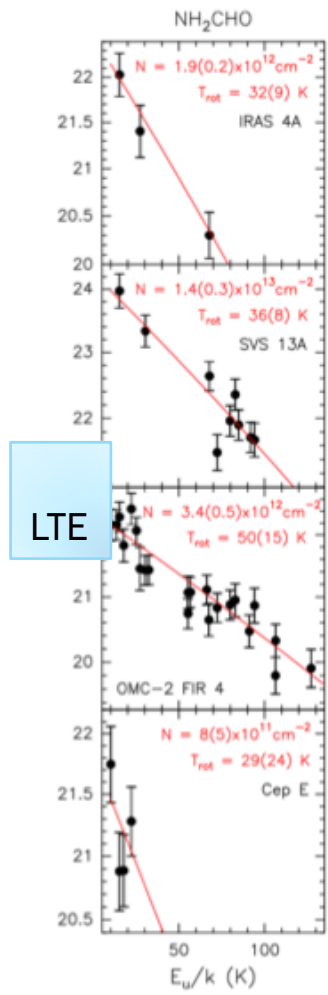
Level population



Example



No LTE

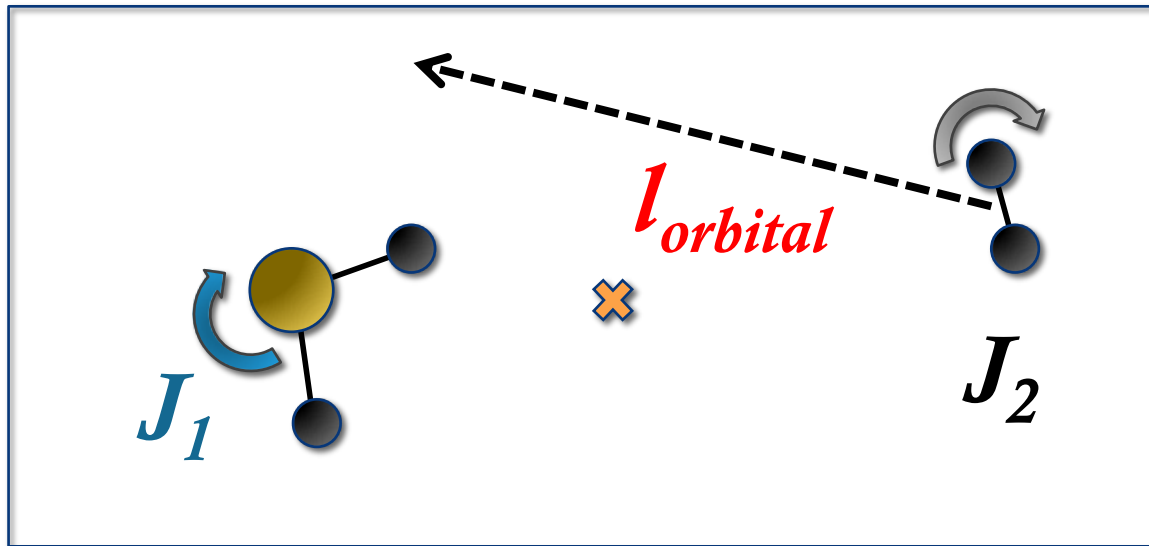


Courtesy of C. Ceccarelli, C. Kahane
 Al-Edhari

Classical view for reaching LTE

- $l_{\text{orbital}} = p b = m v b$

(m reduced mass; b impact parameter; v relative velocity) . **Exchange $l \leftrightarrow J$ makes LTE**





- Full D potential (9D) with H₂, rigid potential with He
- Full rotational analysis, Class./Quantum
- Partial vibrational analysis, Class./Quantum dynamics
- Isotopologues, HDO, D₂O, H₂¹⁸O

MANY COLLABORATIONS

* : isotopologs also computed



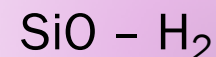
- Potential with CO and H₂ vibration
- High precision rotational analysis

M. Wernli PhD



- Rigid rotor potential with H₂ (4D)
- Full rotational analysis, Quantum dynamics

M. Wernli PhD



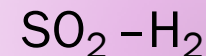
- 5D potential with H₂ & inversion
- All isotopologues NDH₂, NHD₂, ND₃

Several Post-Docs



- Rigid rotor, full potential with H₂
- Overcooled, absorbs the CMB

N. Troscompt PhD



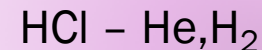
- Full potentiel with He
- Overcooling

Coll. K. Szalewicz



- Full potentiel with He, H₂

E. Sahnoun PhD



- Full potentiel with He, H₂
- Unstable molecule

M. Ben Khalifa PhD



- Full potentiel with He, H₂
- C_{3v} symmetry -> peculiar spectroscopy

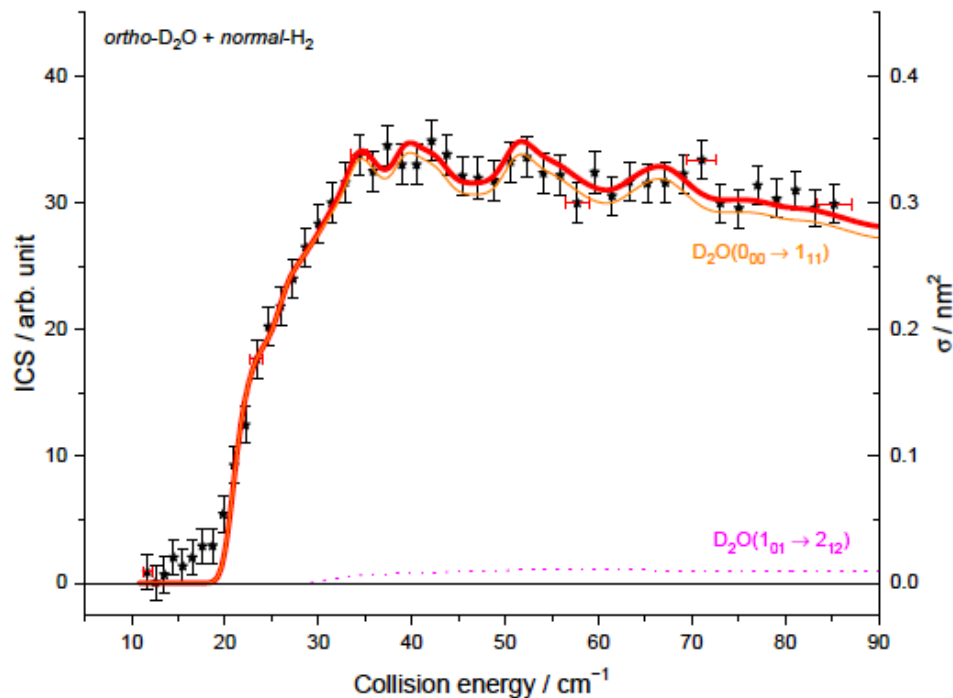
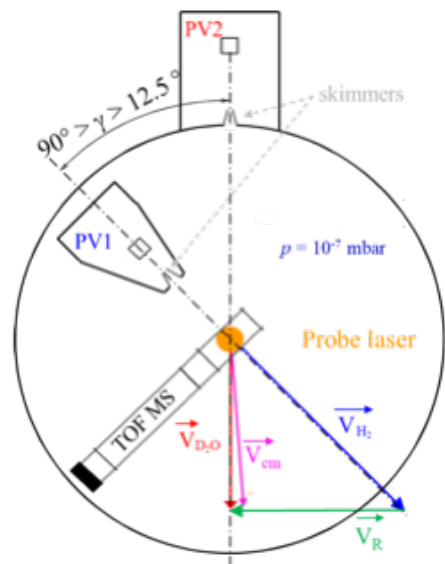
M. Ben Khalifa PhD/post doc



- Interaction with H₂
- Fine structure transition

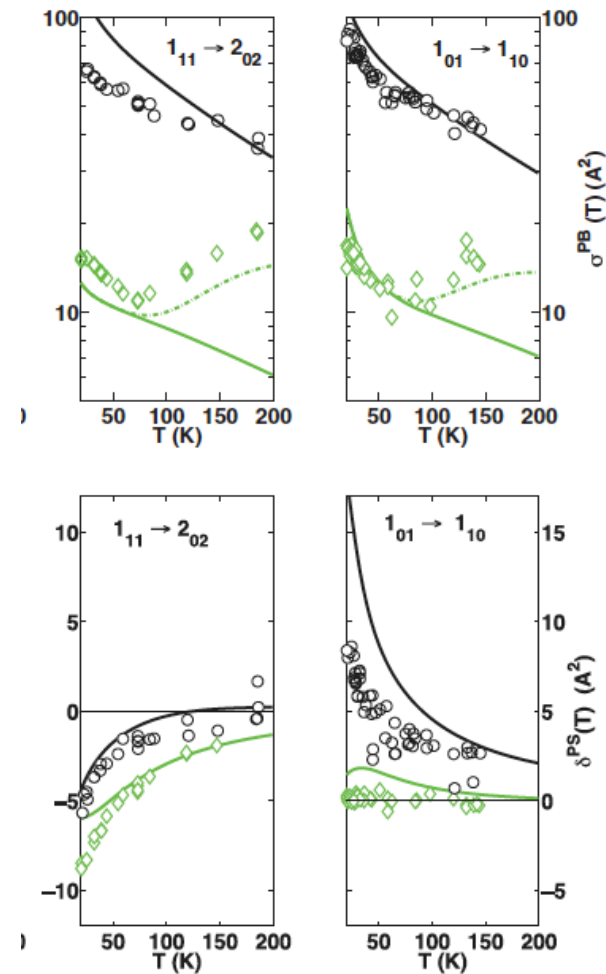
Coll NASA-JPL

Bergeat, LW, Faure, 2020-2022

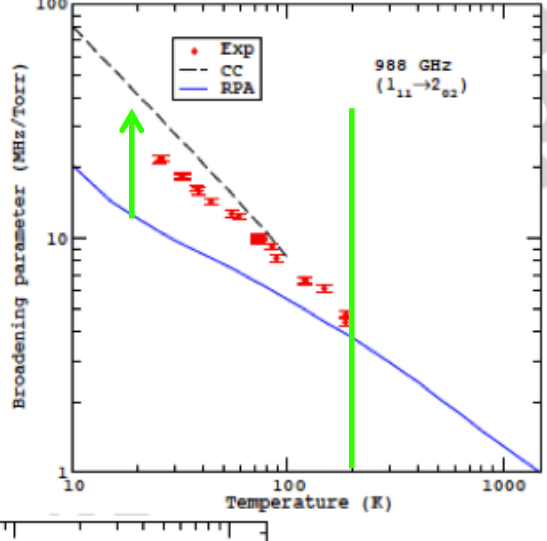
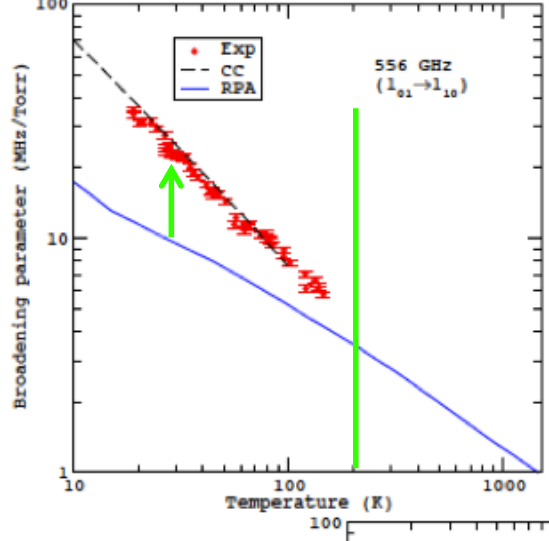
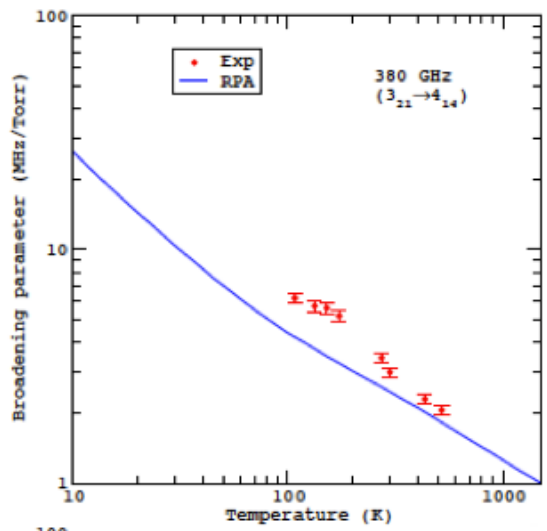
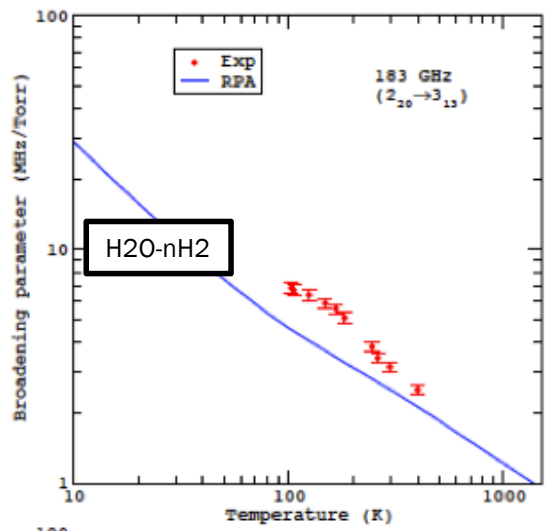


Pressure broadening and shift H₂O – H₂

Drouin, LW, 2012



A. Faure, LW, B Drouin, J Tennyson, JQSRT, 2012.



Moving to atmospheres

CO₂

- PES CO₂ -He CO₂ -H₂
- Dynamics of collision
- Ro-vib in progress
- Pressure Broadening, Opacities

MIT - Harvard - U. Paris
collaboration

CH₄

- PES CH₄ -H₂

Grenoble Tunis
(E. Sahnoun)

Computation on national supercomputers

- **Relevant computations:**
opacity is the MAIN BOTTLENECK for atmospheres.
 - **Quantum mechanical dynamics:**
matrix algebra
-
- Heavy on computation, massively parallel setups.
 - Main project is HPC
 - 10^6 scalar CPU hours is ordinary for a project)
 - For atmospheres (higher T), 10^7 hours projected : large project for present day computing power.
 - MIT-U. Saclay-CNRS joint endeavour being set up

Work in team

- **Lab. Aimé Cotton, Paris-Saclay**

Olivier Dulieu, Alex Voute

- **Grenoble**

Alexandre Faure

- **Tunis**

Nejmeddine Jaïdane

Emna Sahnoun

Malek Ben Khalifa (now, Leuven, BE)

- **MIT / Harvard**

Julie de Wit

Prajwal Niraula

Iouli Gordon

- Many collaborations through **the COST action** « Our astrochemical History »

Extreme case: ultra-small, ultra precise spectroscopic telescope

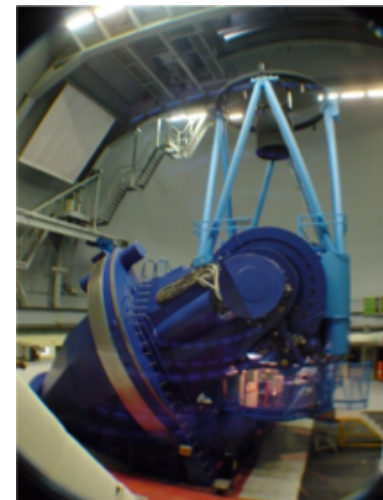
SPECULOOS (Chili- U Liège)



1m telescopes in La Silla and Tenerife

HARPS on the 3.6 m telescope LaSilla

RS=120,000 (measured)



The other extreme: VLT/ESO (Chili)

